

# THE STANDARD FIREWORKS RAJARATNAM COLLEGE FOR WOMEN (AUTONOMOUS), Sivakasi

(Affiliated to Madural Kamaraj University, Reaccredited with "A" Grade by NAAC, College with Potential for Excellence by UGC & Mentor Institution under UGC PARAMARSH)

**NAAC SSR Cycle IV (2015-2020)** 

3.4. PUBLICATIONS

3.4.3. RESEARCH PAPER IN JOURNALS

**EVIDENCES FOR PUBLICATION IN JOURNALS (with DOI Number)** 

2015-2016



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#### **Publication in Journals**

2015-2016

S.No	Name of the author/s	Department of the teacher	Title of paper	Name of journal	Is it listed inUGC CARE/Scopus /Web of Science/other, mention	DOI
1.	Dr.R.Kalaivani	History	A Study on Corporal Punishment in Schools at Sivakasi	International Journal of Research - Granthaalaya h	J- Gate	Nil
2.	Dr.R.Kalaivani	History	Devadasi System in India and Its Legal Initiatives – An Analysis	Journal Of Humanities And Social Science	Index Copernicus	10.979
3.	Dr.T.Palaneeswari	Commerce	Awareness and Satisfaction of Customers of Commercial banks towards Banking Products and Services	Research Journal of Social science and Management	Index Copernicus	Nil
4.	Dr.M.Jayalakshmi And Mrs.K.Sounthara priya	Commerce	An analysis of level of satisfaction towards export of Printing products	International journal of research in commerce, economics and management	SJIFactor	Nil
5.	Dr.J.JeevaPriya	Commerce	Performance Evaluation of MSME under industrial estates programme	International Journal of Research in Social Science	UGC Approved Journal	https://ww w.indianjo urnals.co m/ijor.asp x?target=i jor:ijrss& volume=6 &issue=4 &article= 013
6.	Dr.K.Rajeswari & Dr.K.J.Sunmista	Commerce	Quality of Worklife of Private sector	International Journal of Research in	Scirus	https://sch olar.googl e.com/cita

			bank employees in Sivakasi	Social Science		tions?vie w op=list works&h l=en&user =U- mJzlQAA AAJ
7.	Dr.N.Rajathilagam	Commerce	Analysis of accident causes in fireworks industries by factor analysis technique	International Journal of Management and Social Sciences Research	Corpus ID: 114559536	Nil
8.	Mrs.R.Malini Devi	Mathematics	Analytical Expression for the concentration of Substrate and Product in Immobilized Enzymes system in Biofuel/Biosensor	Applied Mathematics	Web of Science	doi: 10.4236/a m.2015.67 105.
9.	Mrs.SP.Nandhini	Mathematics	R-alpha Open sets in topological spaces	International journal of Mathematical Archive	UGC APPROVED, Index Copernicus	Nil
10.	Mrs.SP.Nandhini	Mathematics	Study on strongly irregular fuzzy graph	International journal of Mathematical Archive	UGC APPROVED, Index Copernicus	Nil
11.	Dr.S.Pethanachi Selvam	Mathematics	Inverse complementary domination graph	International journal of Mathematical trends and technology	UGC CARE List in 2017 and Thomas Reuters	https://doi .org/10.14 445/22315 373/ijmtt- v25p502
12.	Dr.S.Pethanachi Selvam	Mathematics	Line Corporate domination graph	Mathematical Sciences International Research Journal	UGC approved	Nil
13.	Dr.S.Pethanachi Selvam	Mathematics	Enumeration of non- isomorphic semigraphs in Gamma_4	ARS Combinatoria	Scopus	Nil
14.	Dr.S.Pethanachi Selvam	Mathematics	Some results on Detour Radial graph	International journal of Research in Engineering and Applied Sciences	SCIRUS	Nil

15.	Mrs.R.Malini Devi	Mathematics	The Mathematical Theory of Diffusion and Reaction in Enzymes Immobilized Artificial Membrane.	The Journal of Membrane Biology	Scopus	https://doi .org/10.10 07/s00232 -015- 9829-2
16.	Dr.N.Vijaya, Dr.S.Selvalakshmi	Physics	Investigation of Agar-based solid polymer electrolyte	Indian Journal Of Applied Research	Indian Citation Index, Pubmed	Nil
17.	Dr.F.Kingslin Mary Genova, Dr.N.Vijaya and Dr.S.Sivadevi	Physics	Lithium Ion- conducting Blend Polymer Electrolyte Based on PVA–PAN Doped with Lithium Nitrate	Polymer- Plastics Technology and Engineering	Scopus, PubMed, Web of Science	https://doi .org/10.10 80/036025 59.2015.1 050523
18.	Dr.M.Premalatha, Dr.N.Vijaya, Dr.S.Selvalakshmi	Physics	Study of Proton Conducting Blend Polymer Electrolyte Based on PVA-PVP	PARIPEX - Indian Journal Of Research	UGC approved	Nil
19.	Ms.S.Kavitha, Ms. S.Karthieswari, Dr.K. P.Radha	Physics	Effect of PC on Solid Polymer Electrolyte Based on PVA	Global Journal for Research Analysis	UGC Approved, Indian Citation Index	Nil
20.	Dr.F.Kingslin Mary Genova, Dr.N.Vijaya, Dr.S.Sivadevi	Physics	Study on blend polymer (PVA- PAN) doped with lithium bromide	Polymer Science Series A	UGC-CARE List, Scopus	https://doi .org/10.11 34/S0965 545X1507 0032
21.	Dr.K.P.Radha	Physics	AC Impedance And Vibrational Studies Of Biopolymer Electrolyte Cornstarch: NH4SCN	International Education and Research Journal	Index Copernicus, Pubmed, Indian citation index	Nil
22.	Dr.K.P.Radha	Physics	Magnitude Bode Plot Analysis of Solid Polymer Electrolyte PMMA Complexed with Adipic Acid	Der Pharma Chemica	Scopus, DOAJ	Nil
23.	Dr.K.P.Radha	Physics	Admittance Analysis of Proton	PARIPEX - Indian	UGC approved	Nil

			conducting polymer electrolyte PVA:NH 4PF6 :ZrO2	Journal Of Research		
24.	Dr.K.P.Radha	Physics	Effect of Adipic acid on PVA based proton conducting polymer electrolyte	International journal of Applied Research	Scirus, Index Copernicus	Nil
25.	Dr.N.Vijaya	Physics	Preparation and Impedance Analysis of Bio Degradable Polymer Poly Vinyl Alcohol with Amino Acid, Arginine	Polymer- Plastics Technology and Engineering	Scopus, PubMed, Web of Science	https://doi .org/10.10 80/036025 59.2015.1 103263
26.	Dr.N.Vijaya, Dr.S.Selvalakshmi, Dr.M.Premalatha	Physics	Characterization of blend polymer PVA-PVP complexed with ammonium thiocyanate	Ionics	UGC-CARE List, Scopus	https://doi .org/10.10 07/s11581 -016- 1672-7
27.	Dr.K.P.Radha	Physics	Dielectric and ionic conductivity analysis of solid polymer electrolyte based on PMMA	International Journal of Advanced Scientific Research	Index Copernicus	Nil
28.	Dr.F.Kingslin Mary Genova	Physics	Characterization of proton conducting blend polymer electrolyte using PVA-PAN doped with NH <sub>4</sub> SCN	AIP Conference Proceedings	Web of Science, Scopus	https://doi .org/10.10 63/1.4947 864
29.	Dr.S.Selvalakshmi and Dr.N.Vijaya	Physics	Impedance studies of a green blend polymer electrolyte based on PVA and Aloe- vera	AIP Conference Proceedings	Web of Science, Scopus	https://doi .org/10.10 63/1.4948 210
30.	Dr.S.Shanthi	Chemistry	Studies on Photodegradation of Bismarck Brown and Rhodamine B Dye from aqueous	International Journal of Research in Pharmacy and Chemistry	Index Copernicus	Nil

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			solutions of their			
			binary mixture by			
			the synthesized			
			iron oxide nano			
			particles using			
			sunlight and			
			photoreactor			
			Synthesis and			
			characterization of	International	Scopus,Index	
31.	Dr.N.Umasangari	Chemistry	nano sized ZnO	Journal Of	Copernicus	Nil
31.	Di.iv.Oiliasaligaii	Chemistry	using conventional	Chemtech	International,	1111
			and microwave	Research	Scimago.	
			heating methods			
			In silica drug	International		
			activity prediction	Journal of		
32.	Dr.S.Shanthi	Chemistry	of chemical	Engineering	UGC CARE	Nil
		ř	components of	and Applied		
			Acalypha Indica	Sciences		
			Synthesis Spectral			
			Characterization			
22	Dr.P.R.Kavitha	<b>C1</b>	and Molecular	Asian Journal	Mag g PE	N 7'11
33.	Rani	Chemistry	docking studies of	of Chemistry	UGC CARE	Nil
			lawsone as protein			
			kinase inhibitors			
				Indian		
		_	Treatment of	Journal of	UGC CARE	DOI:10.3
34.	Dr.U.Umadevi	Botany	Tetanus in Goats	Applied	list - 49333	6106/ijar
			and Sheep	Research		,
			Successful		т 1	
			treatment of	The Journal	Index	DOI 10.2
35.	Dr.U.Umadevi	Botany	Acaricide	of veterinary	Medicus,	DOI:10.3
		,	Resistant Ticks in	science	Pubmed,	6106
			Cattle		NCBI, DJOF	
			Successful			
			homeopathic	International		
0.5	D IIII 1	D .	treatment of	Journal of	Cross ref,	3.711
36.	Dr.U.Umadevi	Botany	transmissible	Science and	Mendeley	Nil
			tumour in dogs –	Research		
			Case report			
			Influence of			
			Brassinosteroids(	Б		
	M. C.M. 1 1 1 1		BRs) on the	European	G 1	
37.	Ms.S.Muthulakshm	Botany	vincristine content	Journal of	Google	Nil
	i	J	of Catharanthus	Experimental	scholar	
			roseus (L.)	Biology		
			G.Don.			
			Successful	Journal of		
			treatment of	Agriculture	Cross ref, J-	
38.	Dr.U.Umadevi	Botany	recurrent milk	and	Gate	Nil
			fever in field	Veterinary		
			10 voi ili iloid	v etermary		

				Science		
39.	Dr.J.Kasthuri	Botany	Production of PHB by the bacterial isolates of various agroindustrial sites	International Journal of Pharma and Bio Sciences	Elsevier Bibliographic Database(Sco pus and EMBASE) SCImago	DOI: 10.1007/9 78-3-319- 20294- 5_23.
40.	Dr.J.Kasthuri	Botany	Depolymerase and esterase assay in Alcaligens faecalis against the copolymerized Acinetobacter junii CN1 PHBV	International Journal of Recent Scientific Research	Index Copernicus Value: 2016- 81.25 Impact Factor 2017: 7.383 'Thomson Reuters' Researcher ID: K-7356- 2016 NAAS Score: *3.65 (2020)	https://doi .org/10.10 16/j.procs. 2015.03.1 78
41.	Dr.B.Deepa	Botany	Bioinspiredsynthe sis of selenium nanoparticles using flowers of Catharanthus roseus(L.) G.Don.and Peltophorum pterocarpum(DC.) Backer ex Heyne – a comparison	International Journal of ChemTech Research	DOAJ, Scimago, Index Copernicus	Nil
42.	Dr.C.Devi Arockia Vanitha	Computer Science	Real Coded Genetic Algorithm for Development of Optimal G-K Clustering Algorithm	Springer LNCS 8947	Scopus	Nil
43.	Dr.C.Devi Arockia Vanitha	Computer Science	Gene Expression Data Classification using Support Vector Machine and Mutual Information-based Gene Selection	Procedia Computer Science, Elsevier	Scopus	Nil
44.	Dr.A.Bamini	Computer Applications	Comparative Analysis of Clustering in Suer Market Dataset	International Journal of Applied Engineering	Scopus	http://dx.d oi.org/10. 18782/232 0-

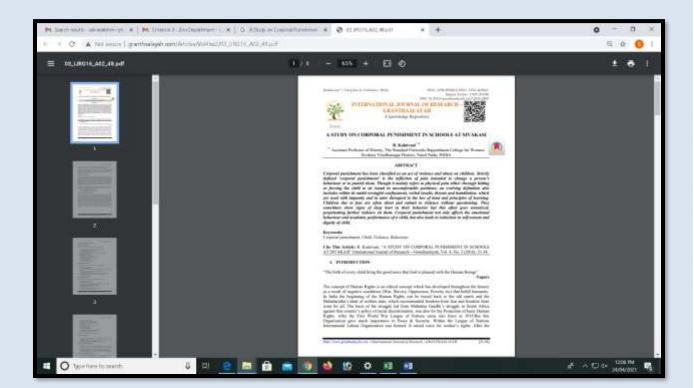
				and Research		7051.2172
45.	Dr.S.Subha Ranjani	Microbiology	Potential influence of probiotic bacteria on the growth gut microflora of Carassius auratus	International Journal of Fisheries and Aquatic Studies	Scirus	Nil
46.	Dr.S.Subha Ranjani	Microbiology	Supplementation of B. cereus as probiotic in fish feed on T. trichopterus (Blue Gourami) and calculating its growth and survival	International journal of current microbiology and applied sciences (NON-UGC)	Scirus	Nil
47.	Mrs.P.Rajeswari	Microbiology	Isolation, Identification and screening of cholesterol degrading probiotics	International journal of Pure and Applied Bioscience (NON-UGC)	ISI	Nil
48.	Mrs.P.Rajeswari	Microbiology	A study on cholesterol degradation by Lactobacillus	Indian journal of Applied Research (NON-UGC)	Pubmed, J- Gate	Nil
49.	Mrs.P.Rajeswari	Microbiology	Invitro study on Hypocholesterole mic effect of spirulina	Indian journal of Applied Research (NON-UGC)	Pubmed, J- Gate	Nil
50.	Dr.S.Radha	Microbiology	Evaluation of probiotic potential of novel candidate Enternacter avium strain from chick faecalborne Alcaligenes faecalis	Global Journal for Research Analysis (NON-UGC)	ICI, Scirus. J-Gate,	Nil



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Name of the Author : Dr.R.Kalaivani

Title of the Paper : A Study on Corporal Punishment in Schools at Sivakasi









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Name of the Author Title of the Paper

: Dr.R.Kalaivani

: Devadasi System in India and Its Legal Initiatives - An

**Analysis** 

IOSR Journal Of Humanities And Social Science (IOSN JHSS) Folume 31. Issue 2. For. II. (Feb. 2013), PP 59-53 e ISSN: 2779-0887, p. ISSN: 2279-0843. www.tostjournals.org

#### Devadasi System in India and Its Legal Initiatives - An Analysis

Mrs. R. Kalaiyani Assistant Professor of History, SFR College For Women, Strahmi

Abstract: The Descripting was once prevailed regist acress halto. It was brown by different names in different places: much as Descripting in Tanell Nada, Maleira in Kerdia, Nata in Aname, Marylli in Mohamatiron, Basanti and Maryala in Aname. Products and Jogatis and Basanti in Karnataka. The word Senadari is the destinal for the country, John and Anamatiron and Anamatiron and International Senadari is the destinated from somethy. The meaning deal and "Basi" maning slower as retrieval name. Bright and the past of Josha, there is an accident retrieval for the past of Josha, there is an accident retrieval for the past of Josha, there is an accident reduction of general grant and the software the meaning to the analysis of the senate in the past of Josha, the class which are the senate in the senate of the senate is defined to the senate for the destination of Konnataka State (South India). These two generations are between as "Devadari" might be committed "Generative the registeration" and they lead a week help. The ward "Devadari" might be committed the senate the senate that the senate is the senate that the senate that the senate is the senate that the senate that the senate is the senate that the senate is the senate that the senate is the senate that the

Keywoods: Devadan, Tellamma, Exploration

I. Introduction

In South India, a devadase (Bassitet: servant of deva (god) or dest (goddess)—is a girl "dedicated" to crossing and service of a desty or a bespite for far over of her life. The dedication takes place in a Furnizative encessive visible is seatled in some ways to manage Congradult, in addition to taking one of the besigle and performing ortals, these visions learned and practiced Ruestmatry, Other and other classical bulks arrests technical one singless a high social metrics at democ and more were seasontial part of temple worthing. Devadual set Men and rossess were dedicated to temple & god's service. They developed the system of muon and dance employed though repair festival. Naturansars were the seals accomposing of service for the performance. They conducted the runter orchestra while the devadase performed her service.

use a pure vicinies. She was so pure, that the could carry visite in a fieldly invalided jet. She would need to use jot everythey on the basils of the viter where the west to first water. On one of these trips, the happened to see a Gausdown'd could be thing in the true. She was trued by the handones before the first male patter as the sure has reflection in the error spatter. But, became of this 'alleged violation' of her manifest over, the pot broke and





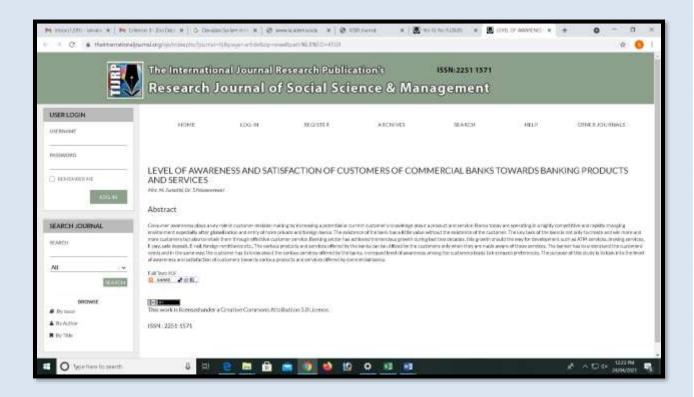
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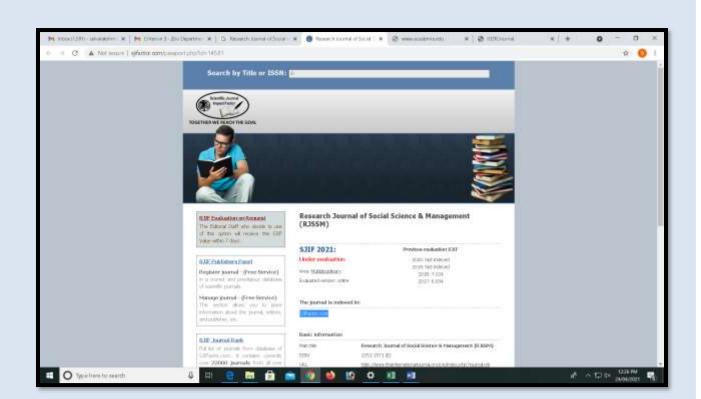
Name of the Author Title of the Paper : Dr.T.Palaneeswari

: Level of Awareness and Satisfaction of Customers of

**Commercial banks towards Banking Products and** 

**Services** 





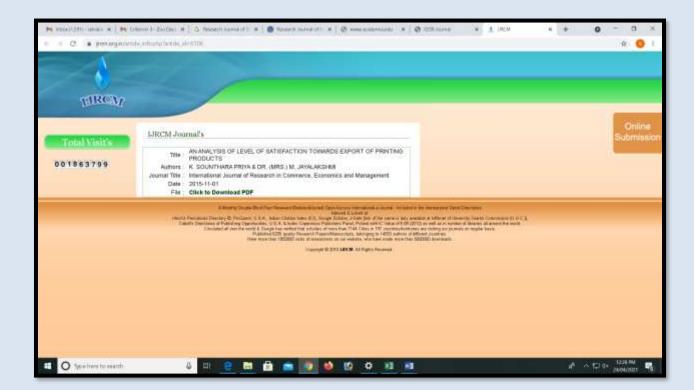


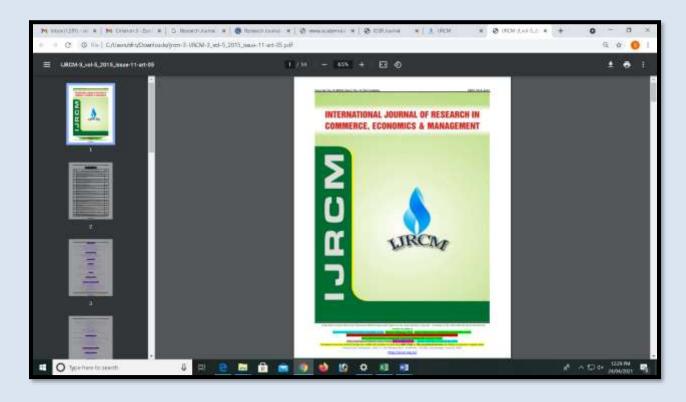
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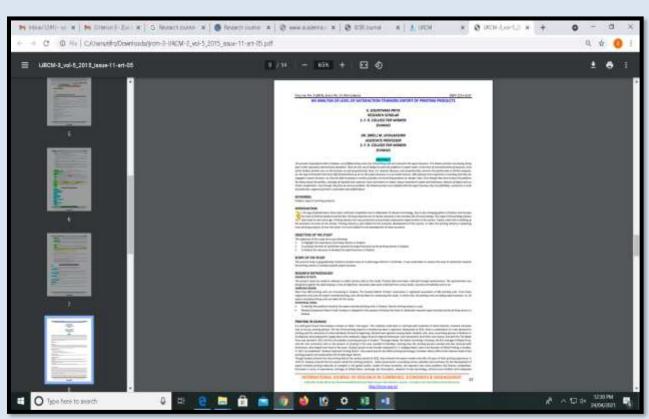
Name of the Author : Dr.M.Jayalakshmi And Mrs.K.Sounthara priya

Title of the Paper : An analysis of level of satisfaction towards export of

**Printing products** 







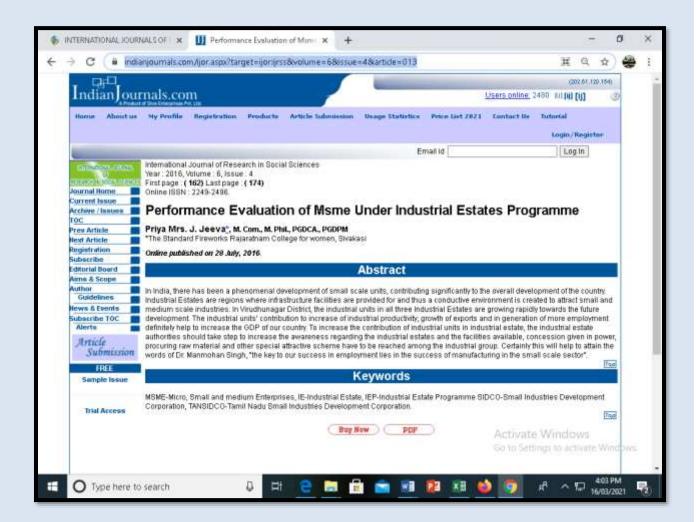


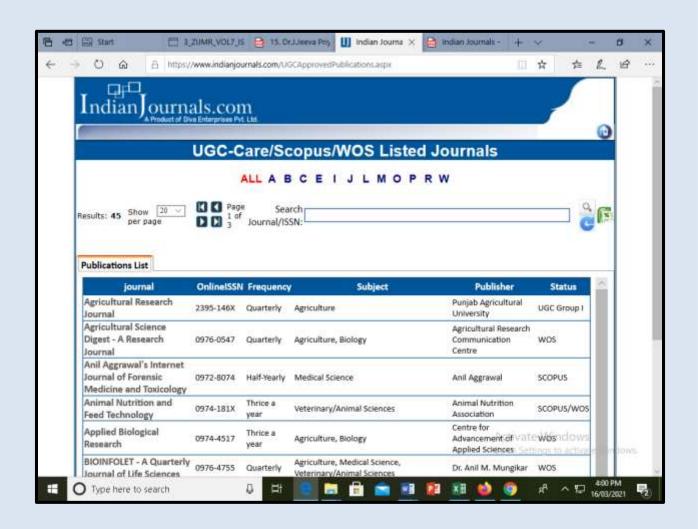
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Name of the Author : Dr.J.JeevaPriya

Title of the Paper : Performance Evaluation of MSME under industrial

estates programme





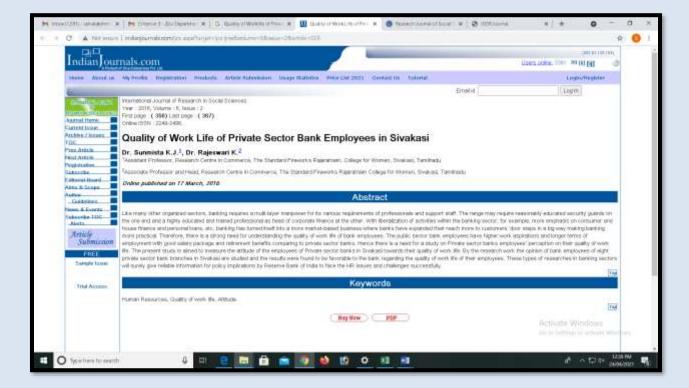


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Name of the Author Title of the Paper : Dr.K.Rajeswari & Dr.K.J.Sunmista

: Quality of Worklife of Private sector bank employees in

**Sivakasi** 





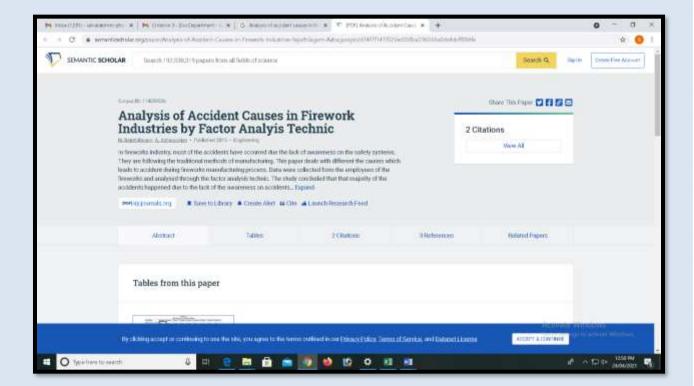


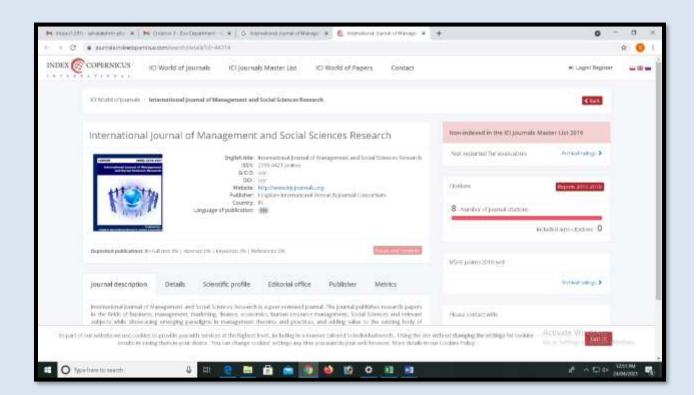
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Name of the Author : Dr.N.Rajathilagam

Title of the Paper : Analysis of accident causes in fireworks industries by

factor analysis technique







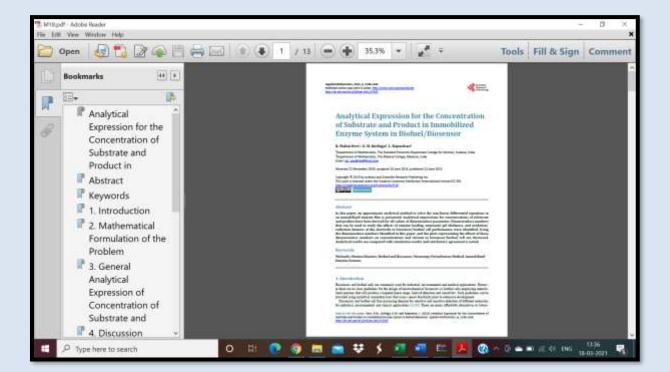
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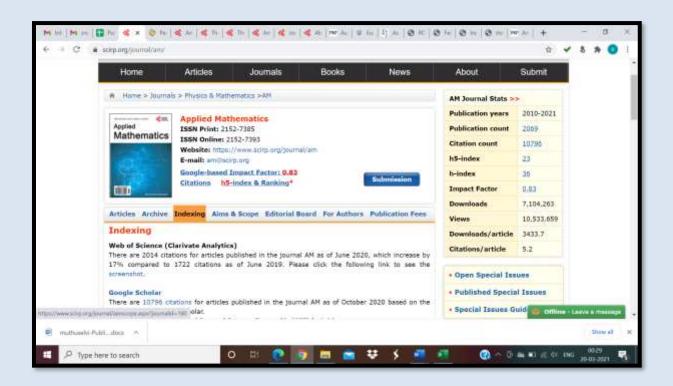
Name of the Author : Mrs.R.Malini Devi

Title of the Paper : Analytical Expression for the concentration of Substrate

and Product in Immobilized Enzymes system in

**Biofuel/Biosensor** 



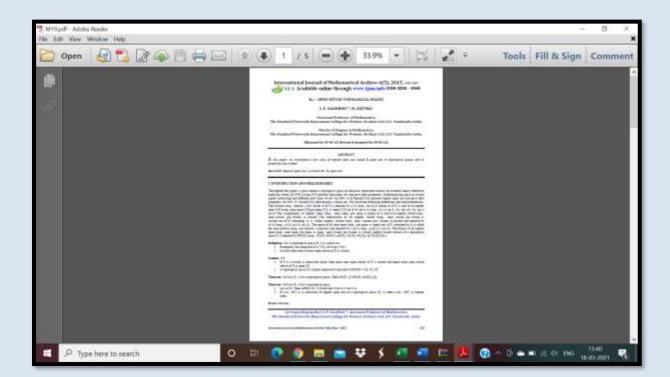


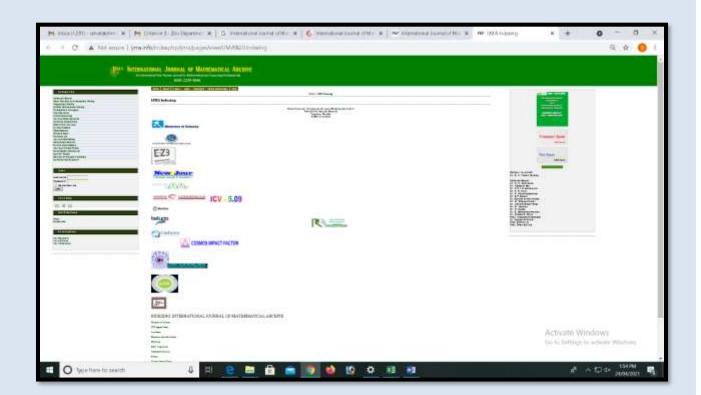


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Name of the Author : Mrs.S.P.Nandhini

Title of the Paper : R-alpha Open sets in topological spaces



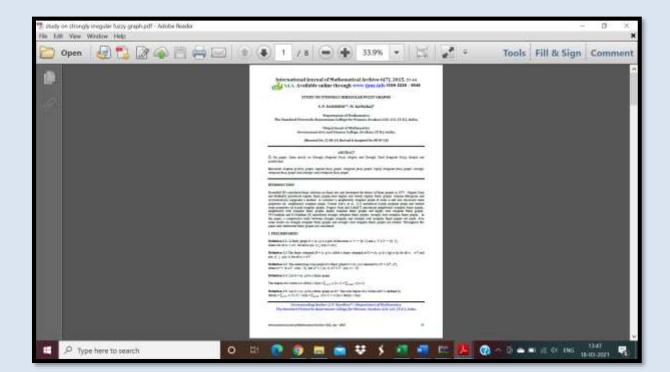


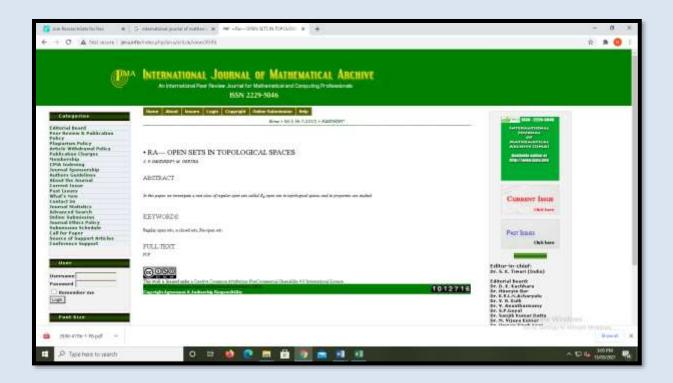


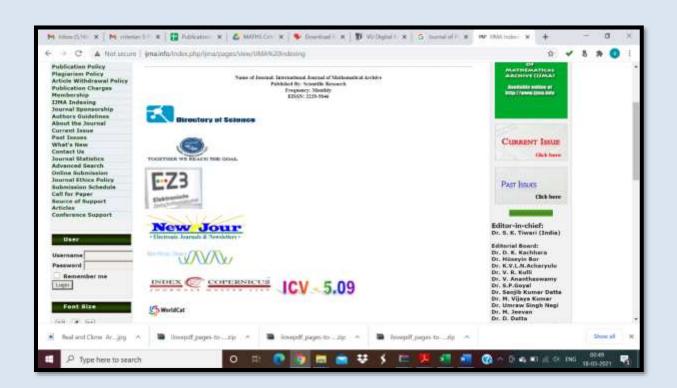
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Name of the Author : Mrs.S.P.Nandhini

Title of the Paper : Study on strongly irregular fuzzy graph





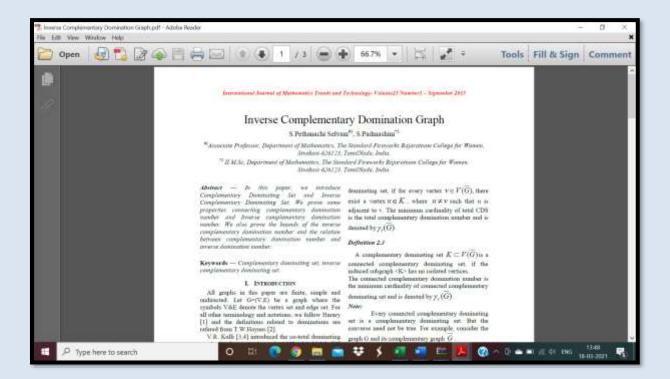




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Name of the Author : Dr.S.Pethanachi Selvam

Title of the Paper : Inverse complementary domination graph



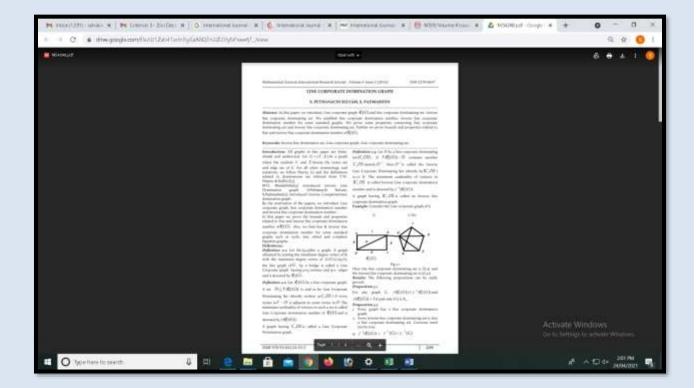


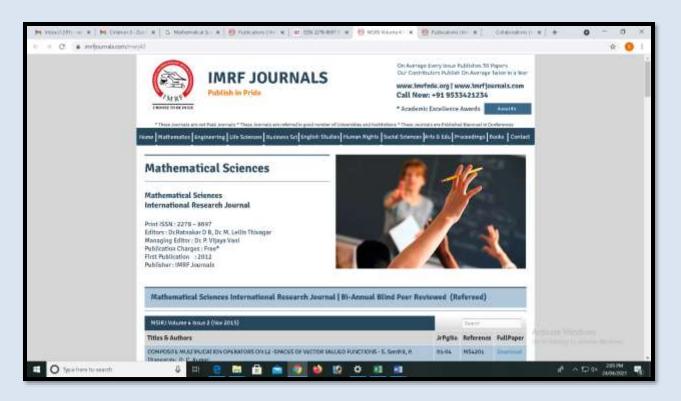


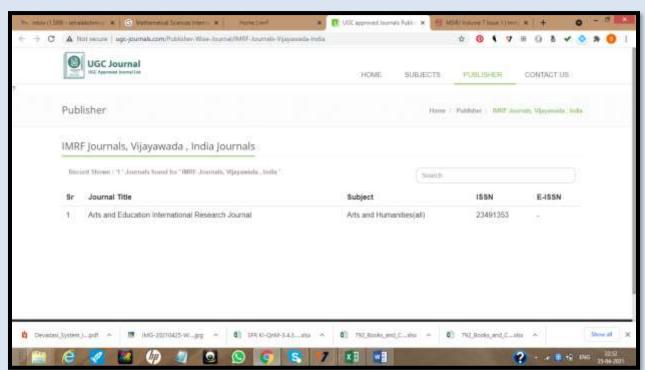
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Name of the Author : Dr.S.Pethanachi Selvam

Title of the Paper : Line Corporate domination graph





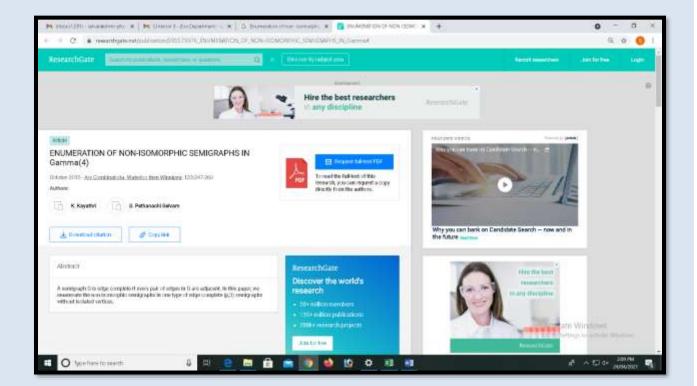




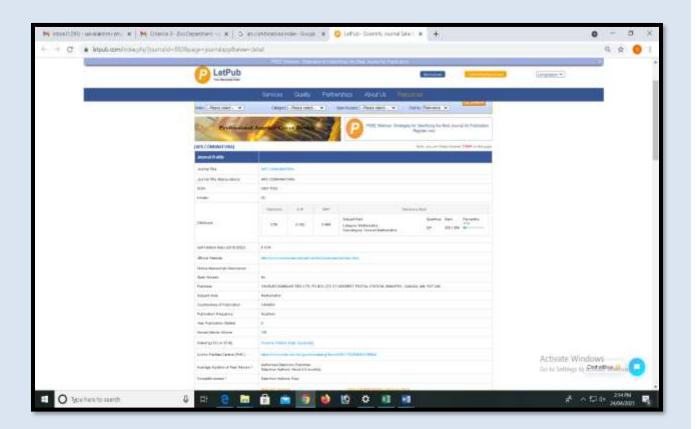
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Name of the Author : Dr.S.Pethanachi Selvam

Title of the Paper : Enumeration of non- isomorphic semigraphs in Gamma\_4





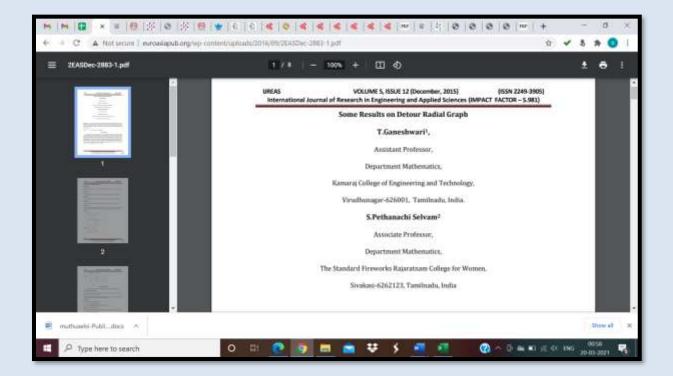


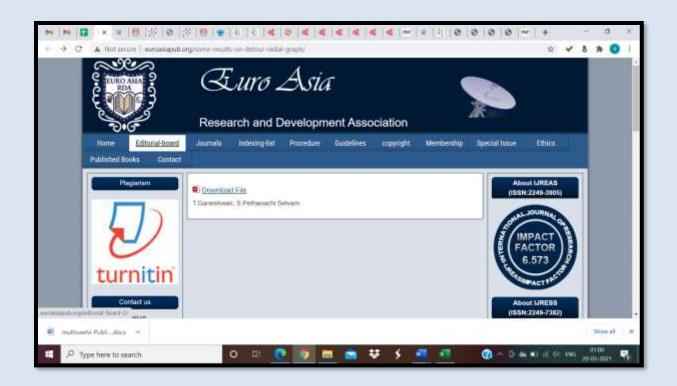


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Name of the Author : Dr.S.Pethanachi Selvam

Title of the Paper : Some results on Detour Radial graph







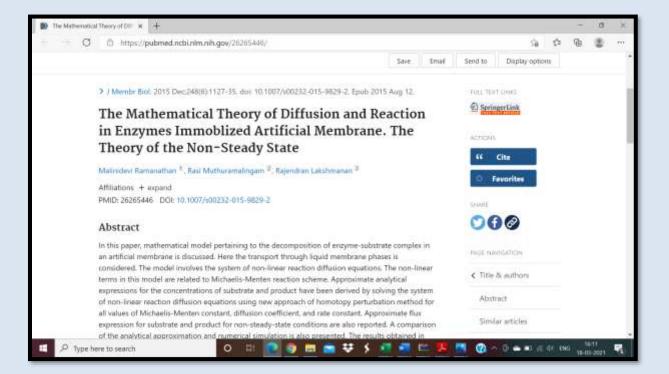


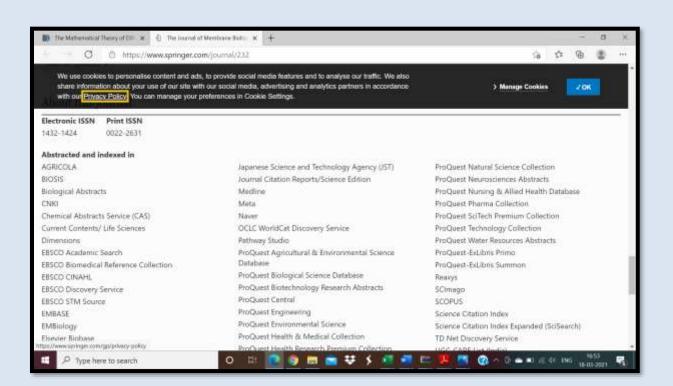
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Name of the Author : Mrs.R.Malini Devi

Title of the Paper : The Mathematical Theory of Diffusion and Reaction in

**Enzymes Immobilized Artificial Membrane.** 







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Name of the Author : Dr.N.Vijaya, Dr.S.Selvalakshmi

Title of the Paper : Investigation of Agar-based solid polymer electrolyte

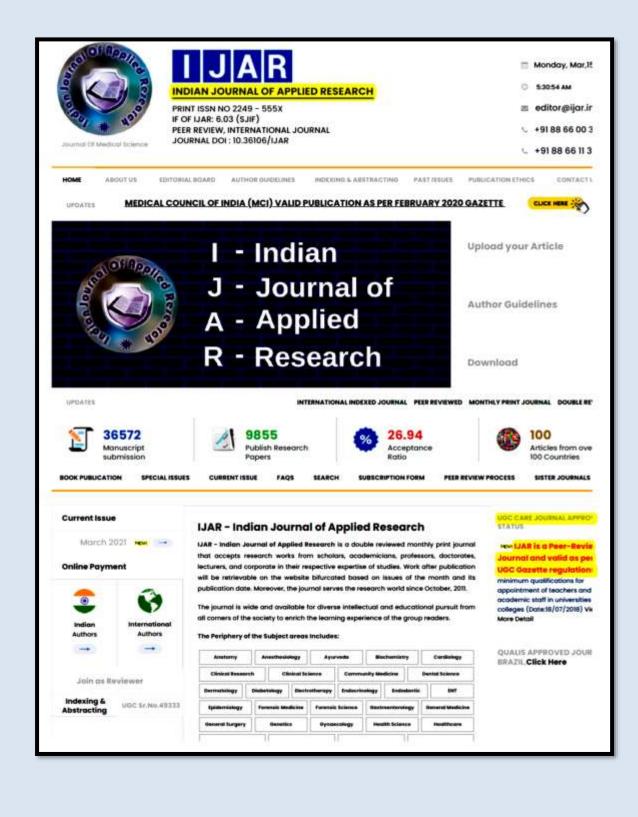


#### 2. Experimental Details:

Agar is insoluble in normal water but readily soluble in boiling water. So Agar-Agar (Colloids Impex Pvt Ltd) of average molecular weight 8000-100000 was dissolved in boiling water and stirred continuously in a magnetic stirrer until a transparent solution was obtained. Then NH, SCN (Reachem) of molecular weight 76.12 was added to the agar solution. The polymer films were prepared with various compositions such as (100.0), (90:10), (80:20), (70:30), (80:40), (50:50) and (40:60) in molar ratios of Agar and NH, SCN. The mixture was stirred for 1 hour to obtain a homogeneous mixture. Then the solutions were casted in polypropylene petri dahes and were dried in

The characteristic peak of pure Agar at 1641 cm<sup>-1</sup> assigned to C=O stretching is shifted to 1637 cm<sup>-1</sup> in 60:40 and 50:50 compositions of NF,SCN doped Agar polymer electrolytes. The characteristic peak of pure Agar at 1477 cm<sub>-1</sub> and 1039 cm<sup>-1</sup> are sasigned to CH<sub>2</sub> bending and acissoning respectively have been shifted to lower wavenumbers at 1421 cm<sup>-1</sup> and 1033 cm<sup>-1</sup> respectively in doped samples.

780 # INDIAN JOURNAL OF APPLIED RESEARCH





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Name of the Author Title of the Paper : Dr.F.Kingslin Mary Genova, Dr.N.Vijaya and Dr.S.Sivadevi

: Lithium Ion-conducting Blend Polymer Electrolyte Based on PVA-PAN Doped with Lithium Nitrate

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### Lithium Ion-Conducting Blend Polymer Electrolyte Based on PVA-PAN Doped with Lithium Nitrate

Kingslin Mary Genova F. abc., S. Selvasekarapandian<sup>c</sup>, S. Karthikeyan<sup>d</sup>, N. Vijaya<sup>b</sup>, S. Sivadevi<sup>b</sup> & C. Sanjeeviraja<sup>e</sup>

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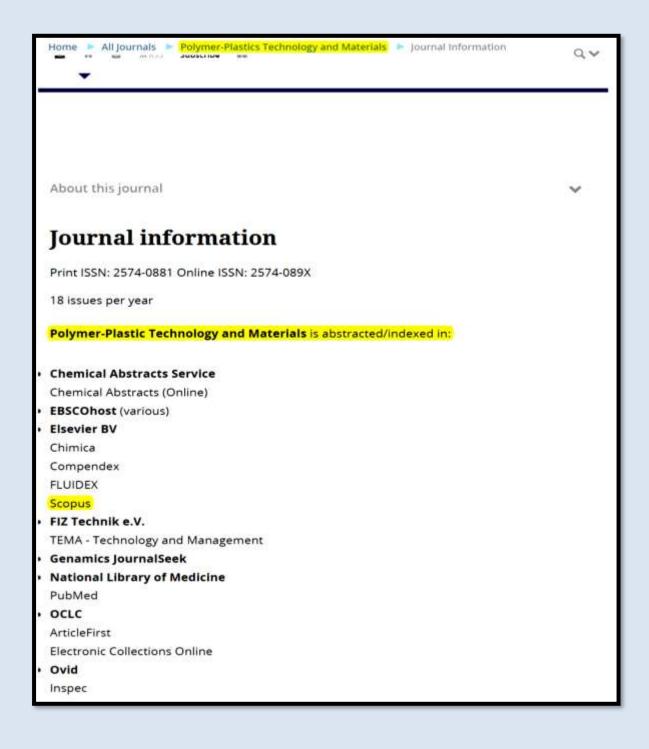
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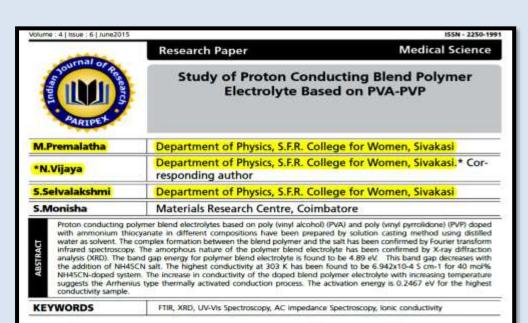
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Name of the Author Title of the Paper

: Dr.M.Premalatha, Dr.N.Vijaya, Dr.S.Selvalakshmi

: Study of Proton Conducting Blend Polymer Electrolyte

**Based on PVA-PVP** 



#### 1. Introduction

1. Introduction
Solid polymer electrolytes play a vital role in the development of electrochemical devices such as batteries, fuel cells, supercapacitors and sensors [1-3]. The main objectives in the polymer research are to develop polymer systems with high conductivity, long life time which are environmentally friendly and have low cost [2,4]. Various approaches have been adopted to improve the electrical, mechanical and electrochemical properties of polymer electrolytes. Polymer blending is one of the most promising ways by which these properties could be changed. Polymer blends are physical mixtures of structurally different polymers that interact through secondary forces and that are miscible to the molecular level. The significant advantages of polymer blends are that the properties of the final product can be tailored to the requirement of applications. product can be tailored to the requirement of applications, which cannot be achieved by one polymer. Generally, blending of two polymers not only results in the improvement of me-chanical strength but also helps increasing the conductivity by suppressing the crystallization of polymer chain [5].

Poly (viryl alcohol) (PVA) is a semicrystalline and biodegrada-ble polymer containing a hydroxyl group attached to methane carbons which can be a source of hydrogen bonding. Poly (viryl pyrrolidone) (PVP) is a synthetic biodegradable polymer and it deserves a unique attention among the conjugated pol-ymers due to its easy processability, moderate electrical con-ductivity and rich in charge transport mechanism. Both these polymers are environmentably friendly and having high polar polymers are environmentally friendly and having high polar

The aim of present work is to develop a proton conducting blend polymer electroyte based on PVA-PVP doped with different concentrations of Ammonium thiocyanate. Ammonium salts have already been reported as good proton donors to the polymer matrix [6,7].

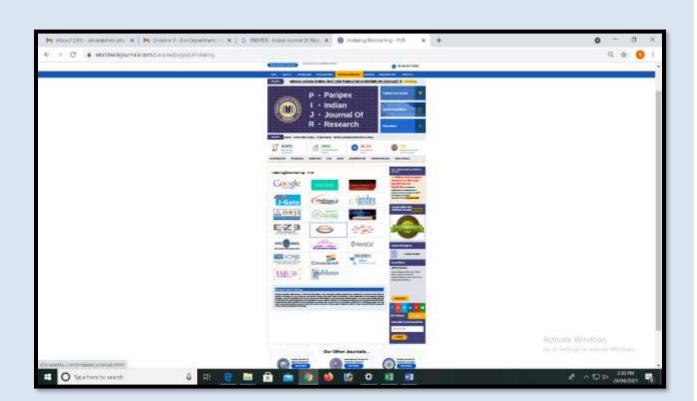
Experimental Technique
In this work, PVA (Mw: 125,000, 5 d fine chem-Ltd, India),
PVP (Mw: 44,000, 5 d fine chem-Ltd, India) and NH<sub>3</sub>SCN
(Reachem) were used as raw materials. Blend polymer elec-

trolytes were prepared with 50 mol% PVA:50 mol% PVP and trolytes were prepared with 50 mol% PVA:50 mol% PVP and ammonium thiocyaniate of various concentrations in mol% using distilled water as solvent by solution casting technique. PVA was stirred in distilled water for 3 hours and after its complete dissolution, PVP was added and stirred for 2 hours after which different concentration of (10%, 20%, 30%, 40%, 50%) ammonium salt was added. The mixture was stirred till it became homogeneous. Then the homogeneous viscous solution was poured in the polypropylene petri dishes and kept in vacuum oven at 60°C for two days to obtain free standing flexible thin films.

A SHIMADZU-IR Affinity-1 spectrophotometer was used to carry out Fourier transform infrared spectroscopy (FTIR) studies in the wave number range of 500-4000 cm<sup>-1</sup>. The optical properties of the films were characterized by UV-VIS spectroscopy. The electrical properties of the blend polymer electrolytes were investigated by using a computer controlled HICKI 3532 LCR analyzer in the frequency range of 42 Hz-1 MHz over the temperature range of 303 K-343 K.

3. Results and discussion
3.1. FTR Characterization
The complex formation between the polymer blend and salt was analyzed by FTIR spectroscopy. The FTIR spectra of pure blend polymer (50% PVA-50% PVP) and the blend polymer doped with 40% of NH, SCN is shown in figure 1. The absorption peaks observed in the pure blend 50% PVA-50% PVP are 3379 cm², 2948 cm², 1743 cm², 1650 cm², 1288 cm², 842 cm², These peaks are attributed to 0-H stretching, CH, saymmetric stretching, CH, obstituted in 40% NH, SCN doped sample. The characteristic peaks at 3379 cm² and 2948 cm² are shifted to 3179 cm² and 2958 cm² respectively in the salt doped blend polymer sample. The absorption peaks at 1379 cm² and 2958 cm² and 1650 cm² are displaced to 1752 cm² and 1639 cm² mespectively in " are displaced to 1752 cm " and 1639 cm" respectively in the blend polymer with 40% NH<sub>2</sub>SCN. The peak at 1288 cm<sup>-1</sup> gets shifted to higher wavenumber at 1292 cm<sup>-1</sup> in the doped blend film. There is a new peak appearing at 2098 cm<sup>-1</sup> in the

189 | PARIPEX - INDIAN JOURNAL OF RESEARCH





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Name of the Author Title of the Paper

: Dr.K.P.Radha

: AC Impedance And Vibrational Studies Of Biopolymer

**Electrolyte Cornstarch: NH4SCN** 

earch Paper | Chemical Science

E-ISSN: 2454-9916 | Volume: 2 | Issue: 1 | Jan 2016



## AC IMPEDANCE AND VIBRATIONAL STUDIES OF BIOPOLYMER ELECTROLYTE CORNSTARCH:NH,SCN

#### G.GURULAKSHMI' | P. MAHALAKSHMI' | K.P.RADHA' | MONISHA'

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- Materials Research Centre, Coimbatore, Tamilnadu

#### ABSTRACT

Solid polymer electrolytes (SPEs) based on himpolymer play a vital role in creating materials for energy storage devices such as batteries and fuel cells. An attempt has been made to prepare solid polymer electrolyte using the biopolymer starch and the doping salt ammonium thiocyanate (NH48CN) by solution conting technique with DMSO as solvent. The FTIR analysis confirms the complex formation between the polymer and salt. The AC impedance spectroscopy reveals that the biopolymer electrolyte 75 mol?s of cornatarch: 25 mol?s of NH48CN has the maximum ionic conductivity 2.76x10° Sem¹ with low activation energy (0.32eV) among the prepared samples The ionic conductivity of the best conducting electrolyte increases with increase in temperature suggesting that the ionic conductivity is temperature of the first property of the prepared samples The ionic conductivity is temperature. ture dependent. The transport number measurement studies confirm that the ionic conductivity in these polymers is due to proton ions.

KEYWORDS: FTIR, AC impedance, Transport number.

Researches have utilized different starches like arrowroot, corn Researches have utilized different starthes like arrowcot, corn and potato starches. Corn starch is most popular polymer since it is renewable and bio degradable polymer. The pulymer slectrolyte based on potato starch doped with ammonium flotide has the highest ambient temperature conductivity as 2.49x10.75cm.

[Kumar et.al., 2012]. The aim of this work is to develop a proton conducting biopolymer electrolyte based on corn starch doped with ammonium thiocyanate. The prepared electrolytes have been characterized by FTIR analysis and electrochemical impedance spectroscopy in order to analyse the interactions between the polymer host (corn starch) and the charge carriers as well as the influence of the charge carriers on the ionic conductivity and onduction mechanism.

B. Preparation of Samples: Cornstarch (AR grade, Sd fine chem) and NH, SCN purchased from Spectrum have been used as starting materials to prepare proton conducting biopolymer electrolytes. Cornstarch: NH,SCN based biopolymer electrolytes of various melar ratios such as (109-0), (85: 15), (75: 25), and (65: 35) have been prepared by the solution casting technique. Appropriate quantity of Cornstarch and NH,3CN are dissolved in the solvent dimethyl sulphoxide (DMSO) seoparately. Then these solutions are mixed together and stirred well in get homogeneous mixture. The resulting solution is poured on to glass petridishes and is allowed to vacuum dried in the vacuum even for 5 days at 70°C, in order to remove the solvent. The smooth uniform flexible polymer films which are transparent to visible light have been obtained:

#### FTIR studies

PTIR spectra have been recorded in the range of 4000 – 400 cm using SHIMADZU IB Affinity 1 spectrophotometer to identify the implexation behavior of the prepared polymer electrolytes.

#### Conductivity measurements

AC conductivity measurements have been carried out on Cornstarch - NH,SCN systems of uniform thickness having an

area of 1 cm". Polymer electrolytes have been sandwiched between two stainless steel (SS) electrodes applying a potential of 1V from 42 Hz to 1 MHz using HIOKI make LCZ meter (model 3532) interfaced to a computer. The conductivity has been calculated from complex impedance plots of measured impedance (Z) and phase angle (6). The temperature of the cell has been controlled using a thermostat and electrical measurements of the polymer electrolytes have been carried out in the temperature range 303K

## C. RESULTS AND DISCUSSIONS C.1 FTIR studies;

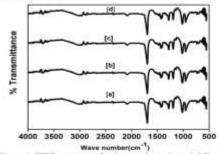
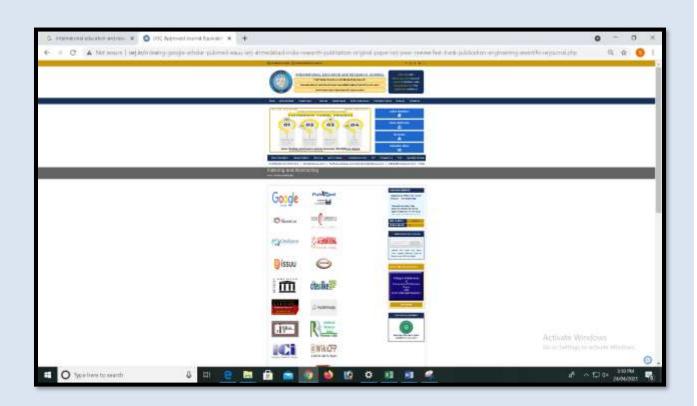


Figure 1: FTIR patterns of polymer electrolytes [a] Pure cornstarch [b] 83Cornstarch:15NH,SCN [c] 75Cornstarch:25NH,SCN [d] 65Cornstarch:25NH,SCN

The PTIR spectra of pure cornstarch and different molar ratios of NH,SCN doped cornstarch complexes are shown in Figure 1. Th hydroxyl band in the starch film appears at 1699 cm<sup>2</sup>. Th

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Name of the Author : Dr.K.P.Radha

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#### Magnitude Bode Plot Analysis of Solid Polymer Electrolyte PMMA Complexed with Adipic Acid

K. P. Radha\*, P. Mahalakshmi and S.Chitra

Department of Physics, S.F.R. College for Women, Sivakasi, Tamilnadu, India

#### ABSTRACT

In the present work, proton conducting solid polymer electrolyte have been prepared using PMMA and adipic acid  $(C_0H_{10}O_0)$  by Solution Casting Technique. From the Magnitude Bode Plot, it has been observed that the Ohmic resistance dominates the impedance at the highest frequencies and it can be read from the high frequency horizontal plateau region. Argand plots of both pure PMMA and adipic acid added electrolyte are incomplete half semicircle suggesting the Non - Debye nature of the polymer electrolytes. At higher frequencies due to the periodic reversal of the electric field there is no excess ion diffusion in the field direction. The presence of long tail in the modulus plot may be due to the large capacitance associated with the electrodes.

Key - words: Argand, Dielectric, Modulus plots

#### INTRODUCTION

Polymer electrolytes combine ionic conductivity in the solid state with mechanical flexibility, making them ideal replacement for liquid electrolytes in electrochemical cells such as Fuel cell, electro chromic displays, sensors etc., because of their ability to form good interfaces with solid electrodes. PMMA is a polymer which forms complexes with inorganic salts such as NH<sub>4</sub>PF<sub>6</sub> LiCF3SO3, LiNO<sub>3</sub> etc.[1]. It has excellent chemical and physical properties which has made it of great interest in proton conducting solid polymer electrolytes. In the present work, we report the effect of adipic acid on the characterization of conventional proton conducting solid polymer electrolytes based on PMMA in terms of dielectric behavior, ionic conductivity and the interactions that had occurred between them.

#### MATERIALS AND METHODS

#### 2.1 Sample Preparation

Polymer electrolytes have been prepared using PMMA:  $C_0H_{10}O_4$  in different molar ratios (100:0), (80:20) by solution casting technique with Dimethyl Formamide (DMF) as the solvent PMMA is stirred in DMF at 24 hours and then  $C_0H_{10}O_4$  is added and stirred at  $55^{\circ}C$  for 12 hours until the mixture become homogeneous viscous liquid. These solutions of different compositions have been poured into identical Petri dishes of 10 cm diameter and are dried in vacuum oven at room temperature for 24 hours. Free standing nature of the electrolyte has obtained.



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: Dr.K.P.Radha

: Admittance Analysis of Proton conducting polymer

electrolyte PVA:NH 4PF6:ZrO2



#### ISSN - 2250-1991 | IF: 5.215 | IC Value: 77.65

**Physics** 

#### Research Paper



Admittance Analysis of Proton conducting polymer electrolyte PVA:NH, PF,:ZrO,

#### Dr. (Mrs) K.P.Radha

#### Associate Professor of Physics, S.F.R. College for Women, Sivakasi, Tamilnadu.



Proton conducting composite polymer electrolyte based on Poly vinyl alcohol (PVA) and ammonium hexafluoro phosphate (NH<sub>2</sub>PF<sub>2</sub>) complex with different molar ratios of the nanofiller Zirconium di oxide. ZrO<sub>2</sub> have been prepared by Solution Casting technique. Concentration dependent Conductivity reveals that the polymer electrolyte 70PVA:30NH<sub>2</sub>PF<sub>6</sub>:22rO<sub>2</sub> has the maximum ionic conductivity (2.5688x10° Scm\*) with minimum activation energy (0.123883 eV) at ambient temperature. The spectroscopic plot of Y" represents the bulk element in series with electrode capacitance. The frequency independent region in the spectroscopic plot of Y' indicates the accumulation of the charge carriers near the electrode.

#### KEYWORDS

AC impedance analysis, admittance, dielectric

Introduction: One way of enhancing the conductivity of polymer electrolytes without significantly compromising their mechanical properties is via the incorporation of the nanofflier ZrO, TiD, AL,O, SiO, or solid plasticizers. These plasticizers impart the dual advantages of increasing local segmental mobility in the polymer chains reflected in decreasing glass transition temperature (Tg) and increasing the degree of dissociation of the salt [1]. When the delectric constant of the solid plasticizer is greater than that of the polymer host the addition of such plasticizer will increase the permittivity of the mature. Effectively solid plasticizer was serve to promote both the number of available charge carriers and their mobility through the polymer network.

In our earlier work, we have deaft with preparation of polymer electrolyte with PVA and ammonium hexafluoro phosphate (NH,PF). In the present work, the optimized high conductivity polymer electrolyte 70PVA:20 NH,PF, (mol %) has been further optimized to find the effect of the nanofiller ZrO, of 45nm size on the ionic conductivity of the proton conducting polymer electrolyte.

## Preparation

Preparation

Preparation

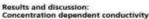
Poly (vrryl alcohol) (PVA) with molecular weight 1,25,000 (AR grade Sd fine chem. make), ammonium hexafluoro phosphate (NH\_PF\_) purchased from Aldrich, USA and the nano filler Zirconium di Oxide (ZrO\_) from Aldrich USA of particle size 4Snm and Dimethyl Sulphoxide (DMSO) as solvent are used as starting material to prepare composite polymer electrolytes by solution casting technique. From our earlier work, it has been observed the optimum concentration of PVA and NH\_PF\_ as 70mol% and 30mol%. The nano filler ZrO\_ is added to this optimum concentration (70PVA:30 NH\_PF\_) as 1mol%, 2mol% and 3mol%. Appropriate weights of PVA, NH\_PF\_ are dissolved in DMSO by using magnetic stirrer. The Nano filler ZrO\_ is suspended in the solution and then stirred well to get nomogenous moture. The mature is then pouned into glass Petri dish and is allowed to evaporate the solvent in the vacurouser at 80°C for 5 days, Free standing nature of the electrolyte has obtained.

#### nductivity measurements:

Conductivity measurements:

AC conductivity measurements have been carried out on PVA - NH PF, systems of uniform thickness having an area of cm? Polymor electrolytes have been sandwiched between two stainless steel (SS) electrodes applying a potential of TV from 42 Hz to 1 MHz using HIOSI make LSZ meter (model 3532) interfaced to a computer. The conductivity has been

calculated from complex impedance plots of measured imped-ance (2) and phase angle (a). The temperature of the cell has been controlled using a thermostat and electrical measure-ments of the polymer electrolytes have been carried out in the temperature range 303K - 343K.



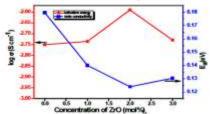


Figure 1 Variation of conductivity and activation energy of 70 PVA: 30 NH<sub>4</sub>PF<sub>4</sub> as a function of ZrO<sub>2</sub> concentration at 303K.

The activation energy of the composite polymer electrolyte decreases with increase in the ionic conductivity as shown in Fig. 1. Since the proton ion transfer is greatly affected by the polymer segmental motion, a composite electrolyte with lower activation energy suggests rapid ion conduction and hence higher conductivity (2). Normally ionic conductivity of the polymer segments are suggested as a conductivity of the polymer conductivity (2). ymer electrolyte depends upon both charge carrier conce tion, n and carrier mobility  $\mu$  as

where q representing the charge of mobile carrier. The charge carrier concentration in depends upon both the dissociation energy U involved and dielectric constant  $\epsilon$  as

 $n = n \exp(-U k \cdot kT)$ 

Where k is Boltymann constant and T the absolute temperature. Upon incorporation of the nanofiller ZPO<sub>2</sub> to the electricitie 70 PVA:30 NH<sub>2</sub>FF<sub>2</sub> dielectric constant has been experimentally found to increase during measurements (Fig.2). It





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: Dr.K.P.Radha

: Effect of Adipic acid on PVA based proton conducting

polymer electrolyte



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Materials Research Centre Compatore Taminado India

#### Effect of Adipic acid on PVA based proton conducting polymer electrolyte

P Mahalakshmi, G Gurulakshmi, Dr. KP Radha, S Monisha

The proton conducting solid polymer electrolytes composed of Poly vinyl Alcohol (PVA) with different concentration of adipic acid (C<sub>8</sub>H<sub>B</sub>O<sub>8</sub>) have been developed by using solution casting technique. The prepared polymer electrolytes have been subjected to XRD, DSC, AC impedance and Wanger's dc Polarization studies. XRD analysis confirms amorphous nature of polymer electrolytes. Wanger's de Polarization shuhes. XRD analysis confarms anioephicus nature of polymer electrolytes. DSC analysis reveals that the best ion conducting sample 75 PVA: 25 CaHis/Ox polymer describylyte has more thermal stability than pure PVA polymer electrolyte. The highest ionic conductivity has been found to be 7.50×10<sup>-7</sup> Scm<sup>-7</sup> at 3/03 K for 75 PVA: 25 CaHis/Ox polymer electrolyte. Temperature dependent conductivity of polymer electrolytes obeys Arrhenius equation. The highest conductivity polymer electrolyte 75 PVA: 25 CaHis/Ox has low Activation energy 0.2744eV among the prepared polymer electrolytes. The transference number of electrolyte in the highest ion conductivity sample is Wanger's dc Polarization analysis suggests that the prepared polymer electrolytes are the best

Keywords: XRD, DSC, AC impedance. Wanger's dc Polarization

#### 1. Introduction

In recent years there has been much interest in ion conducting polymers which show high ionic conductivity at room temperature because these are the most promising electrolytes in various electrochemical devices such as all solid-state batteries, sensors, fuel cells etc. Poly (vinyl alcohol) is semicrystalline polymer and cost effective biodegradable synthetic polymer with high tensile strength and flexibility. It contains hydroxyl group attached to methane carbons which acts as a hydrogen bonding source. Adapte acid  $(C_6H_{10}O_4)$  is a white crystalline powder. This organic compound is the most important synthetic dicarboxylic acid. It has a wide range of applications like coatings, plasticizers and detergents. Adipic acid is readily biodegradable and has a low potential for bioaccumulation.

The intention of the present work is to prepare and characterize polymer electrolytes based on PVA with different compositions of  $C_0H_{10}O_4$ . The prepared samples have been subjected to X-Ray Diffraction, Differential Scanning Calorimetry. Ac impedance Spectroscopy and Wagner's de polarization technique.

## **Experimental Technique**

#### 2.1 Sample Preparation

PVA with average molecular weight 125,000 (AR grade, sd fine chem) and adipic acid  $(C_6H_{10}O_4)$  have been used as starting materials to prepare proton conducting polymer electrolytes. PVA based solid polymer electrolytes of various molar ratios of adipic acid such as (100.0), (80.20), (75.25), (70.30) have been prepared by the solution casting technique. Appropriate quantity of PVA and adipic acid are dissolved in the solvent dimethyl sulphoxide (DMSO) separately. Then these solutions are mixed together and stirred well to get homogeneous mixture. The resulting solution is poured on to glass petri dishes and is allowed to vacuum dried in the vacuum oven for 5 days at 70 °C, in order to remove the solvent. The smooth uniform flexible polymer films which are transparent to visible light have been obtained.

#### Dr. KP Radha

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Name of the Author : Dr.N.Vijaya

Title of the Paper : Preparation and Impedance Analysis of Bio Degradable Polymer Poly Vinyl Alcohol with Amino Acid, Arginine



# Preparation and Impedance Analysis of Bio Degradable Polymer Poly Vinyl Alcohol with Amino Acid, Arginine

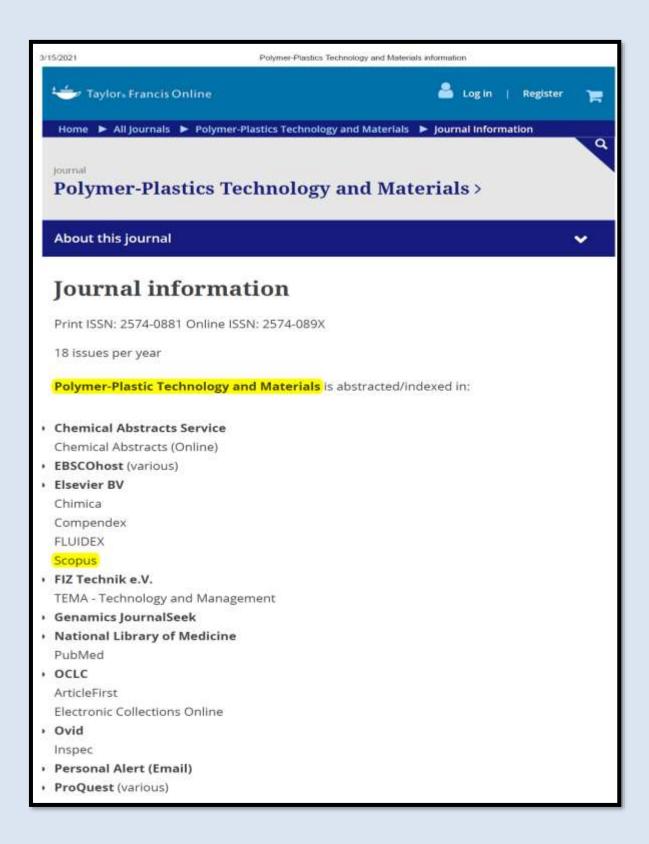
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# Abstract

Bio degradable polymers have an innumerable use in the field of biomedicine, especially in drug delivery system. PolyvinyFalcohol(PVA) is one of the bio degradable polymer, used as a carrier for drug delivery. Amino acids are necessary for maintaining good health for human being. The present study focuses on the interaction between PVA and amino acids. An effort is being taken to prepare polymer membrane based on PVA complexed with different concentration of Arginine, a type of amino acids using water as solvent by solution casting technique. The amphorousity and complex formation between PVA and Arignine have been confirmed by XRD, FTIR spectroscopy respecticvely. The thermal behavior of PVA/Arginine complexes has been analyzed by DSC. From AC impedance spectroscopy, Ion transport mechanism has been investigated in detail. By using Almond and West formulisms, the parameter such as ion hopping frequency  $\omega_p$ , has been calculated. The polymer membrane 75Mwt% PVA: 25Mwt% Arginine has the highest ionic conductivity as  $1.97 \times 10^{-6}$  S cm<sup>-1</sup> at ambient temperature.





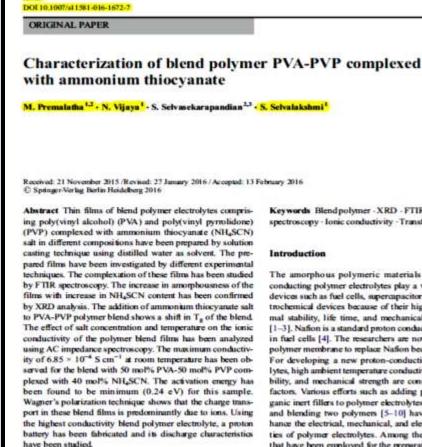
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Name of the Author Title of the Paper

: Dr.N.Vijaya, Dr.S.Selvalakshmi, Dr.M.Premalatha

: Characterization of blend polymer PVA-PVP complexed

with ammonium thiocyanate



Keywords Blendpolymer - XRD - FTIR - AC impedance spectroscopy - Ionic conductivity - Transference number

The amorphous polymeric materials especially protonconducting polymer electrolytes play a vital role in the ionic devices such as fuel cells, supercapacitors, batteries, and electrochemical devices because of their high conductivity, thermal stability, life time, and mechanical strength properties [1-3]. Nation is a standard proton conducting membrane used in fuel cells [4]. The researchers are now developing a new polymer membrane to replace Nation because of its high cost. For developing a new proton-conducting polymer electrolytes, high ambient temperature conductivity, flexibility, durability, and mechanical strength are considered as important factors. Various efforts such as adding plasticizers and inorganic inert fillers to polymer electrolytes, use of copolymers, and blending two polymers [5-10] have been made to enhance the electrical, mechanical, and electrochemical properties of polymer electrolytes. Among the various approaches that have been employed for the preparation of polymer electrolytes with high ionic conductivity and appreciable thermal stability at ambient temperature, blending of polymers is an important method. Polymer blends are polymer systems originated from the physical mixture of two or more polymers and/or copolymers, without a high degree of chemical reactions between them. The advantages of polymer blends lie in the ability to combine existing polymers into new compositions obtaining in this way, materials with specific properties for desired applications. This strategy allows for savings in research and development of new materials with equivalent properties, as well as versatility, simplicity, relatively low cost and faster development time of new materials [11]. The most

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Name of the Author Title of the Paper : Dr.K.P.Radha

: Dielectric and ionic conductivity analysis of solid

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#### Dielectric and ionic conductivity analysis of solid polymer electrolyte based on PMMA

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#### Abstrac

An attempt has been made to prepare a new proton conducting solid polymer electrolyte based on PMMA doped with adipic acid  $(C_6H_{50}O_6)$  by Solution Casting Technique. The complex admittance plot consists of low frequency are and high frequency spike. Arrhenius plot analysis shows that the addition of adipic acid to PMMA enhances the ionic conductivity of Pure PMMA solid polymer electrolyte from  $5.8040 \mathrm{X} 10^{-7} \mathrm{Scm}^{-1}$  to  $1.673 \mathrm{Z} \mathrm{X} 10^{-8} \mathrm{Scm}^{-1}$  at ambient temperature. The curves in the Argand plot at different temperatures are incomplete half semicircle suggesting Non - Debye nature of the electrolytes. The long tail in the low frequency range of M' spectra indicate the capacitive nature of the system.

Keywords: Arrhenius, Argand, Bode plot, Admittance

#### 1. Introduction

Solid Electro Chemical Devices based on Solid Polymer Electrolytes has attracted great scientific interests over the conventional liquid electrolyte due to the problems of leakage. Solid Polymer electrolytes have various advantages such as fast charge transfer at electrode interface. Bexibility, electro-chemical stability, light weighted mechanical strength and good processability. Unfortunately the low conductivity of the solid polymer electrolytes at room temperature limits their practical applications. Thus various efforts have been done to improve the proton conductivity of solid polymer electrolytes. One of the most successful approaches to enhance ionic conductivity is to add salts. Many proton conducting polymer electrolytes such as PVP: NH4SCN, PEO: NH4ClO4. Chitosen: NH4NO3 etc are available in the literature, and they have exhibited excellent proton conductivity. It reveals that animonium salts are very good proton donoss. In this continuation, an attempt has been made to prepare and characterize a new proton conductivity solid polymer electrolyte based on PMMA doped with adaptic acid (CeH10O4).

#### 2. Experimental Procedure

#### 2.1 Sample Preparation

Polymer electrolytes have been prepared using PMMA: ColfiscOa in different molar ratios (100:0), (80:20) by solution casting technique with Dimethyl Formanide (DMF) as the solvent. PMMA is stirred in DMF at 24 hours and then ColfiscOa is added and stirred at 550 C for 12 hours until the mixture become homogeneous viscous liquid. These solutions of different compositions have been poured into identical Petri dishes of 10 cm diameter and are dried in vacuum oven at room temperature for 24 hours. Free standing nature of the electrolyte has obtained.

#### 2.2 Conductivity measurements

AC conductivity measurements have been carried out on PMMA: CaH10Oa systems of uniform thickness having an area of 1 cm<sup>2</sup>. Polymer electrolytes have been sand witched between two stainless steel (SS) electrodes applying a potential of 1V from 42 Hz to 1 MHz using HIOKI make LCZ meter (model 3532) interfaced to a computer. The conductivity have been calculated from complex impedance plots of measured impedance (Z) and phase angle (e). The temperature of the cell has been controlled using a thermostat and electrical measurements of the polymer electrolytes have been carried out in the temperature range 303K-343K.

#### 3. Results and discussion

#### 3.1 Admittance analysis

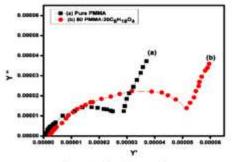
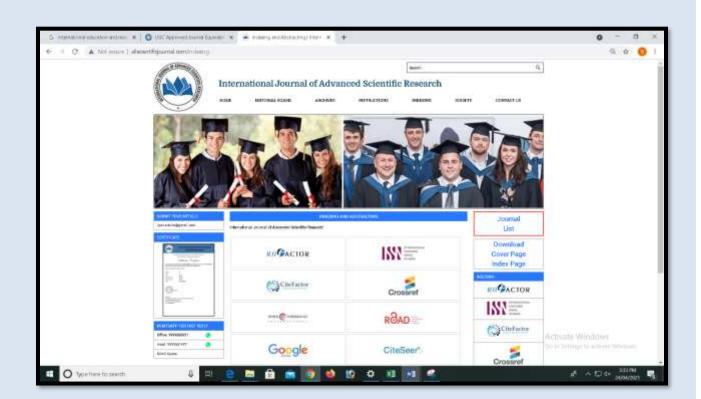


Fig 1: Complex Admittance plot

The complex admittance plot is a powerful method to investigate the ion conduction mechanism. The complex admittance  $Y^*$  is given by the relation

$$Y^* = 1/Z^* + Y^* + jY^*$$
 (1)

Where Y and Y is the real and imaginary part of admittance respectively. Fig 1 [a, b]. Shows Y vs Y of all samples at 303K. The plot shows a low frequency arc and high frequency spike. The spike represents the bulk response. The bulk capacitance ( $C_b$ ) can be calculated at any point along this spike using the relation





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Name of the Author : Dr.F.Kingslin Mary Genova

: Characterization of proton conducting blend polymer Title of the Paper

electrolyte using PVA-PAN doped with NH4SCN

## Characterization of proton conducting blend polymer electrolyte using PVA-PAN doped with NH<sub>4</sub>SCN

M.Premalatha<sup>1,2</sup>, T. Mathayan<sup>1,\*</sup>, S.Selvasekarapandian<sup>2</sup>, F.Kingslin Mary Genova3.\*. R. Umamaheswari3

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Abstract. Polymer electrolytes with proton conductivity based on blend polymer using polyvinyl alcohol (PVA) and poly acrylo nitrile (PAN) doped with ammonium thiocyanate have been prepared by solution casting method using DMF as solvent. The complex formation between the blend polymer and the salt has been confirmed by FTIR Spectroscopy. The amorphous nature of the blend polymer electrolytes have been confirmed by XRD analysis. The highest conductivity at 303 K has been found to be 3.25 x 10<sup>-2</sup> S cm<sup>-1</sup> for 20 mol % NH<sub>4</sub>SCN doped 92.5PVA-7.5PAN system. The increase in conductivity of the doped blend polymer electrolytes with increasing temperature suggests the Arrhenius type thermally activated process. The activation energy is found to be low (0.066 eV) for the highest conductivity sample

Keywords: FTIR, XRD, AC impedance, activation energy

#### INTRODUCTION

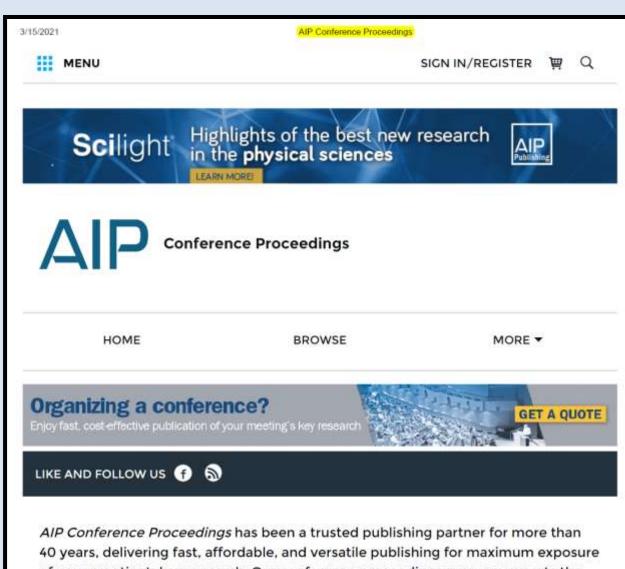
In the field of electrochromic energy devices such as batteries, fuel cells etc., proton conducting polymer electrolytes play a essential role because of its mechanical and electrical properties [1]. Various methods have been adopted to improve the electrical. mechanical and electrochemical properties olymer electrolytes. Among the various methods, Polymer blending is the most promising way to improve these properties which is a mixture of structurally different polymers. There are many reports available based on PVA-PVP [2], P[VdF-HEP]- PVAc[3], PVAc-PMMA[4], PVdF-PMMA[5] and so on. PVA is a semi crystalline polymer containing hydroxyl group attached to methane carbons which can be a source of hydrogen bonding PAN is a synthetic, semi crystalline organic polymer resin which is a common substitute for wool in clothing and home furnishings. It is already reported that the molecular weight of 92.5%PVA: 7.5% PAN exhibits the conductivity of 1.13 × 10<sup>-7</sup> S cm<sup>-1</sup> at room temperature [6]. In this work, proton conducting polymer blend electrolyte based on PVA-PAN doped with ammonium thiocyanate (NH<sub>4</sub>SCN) is synthesized and characterized. Ammonium salts have already been reported as a good proton donors to the polymer matrix [7].

#### EXPERIMENTAL TECHNIQUE

Blend polymer electrolytes are prepared with optimized compositions of 92.5% PVA(MW 1,25,000): 7.5 % PAN (MW 1,40,000) and various compositions of ammonium thiocyanate using dimethyl formamide (DMF) as solvent by solution casting technique, 92.5% weight of PVA is stirred in DMF at 60°C for 3 hours and after its complete dissolution, 7.5 % weight of PAN is added and stirred for 2 hours after which the ammonium salt is added. The mixture is stirred till it becomes homogeneous. Then it is poured in the petri dish and evaporated at 60°C in vacuum oven. Free standing film is obtained after 48 hours. Films obtained with 5 mol%, 10mol %, 15 mol%, 20mol% and 25mol % by weight of ammonium thiocyanate. Then the film is carefully removed from the petridish and sealed in an airtight cover. Then the prepared films were subjected to Fourier transform infrared spectroscopy (FTIR) using SHIMADZU-IR Affinity-1 spectrophotometer. The XRD patterns were recorded at room temperature using a XPERT-PRO Diffractometer with Cuko radiation at 40 KV and 30 mA in the 20 range of 10 -80'. The electrical properties of the films were

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Name of the Author : Dr.S.Selvalakshmi and Dr.N.Vijaya

Title of the Paper : Impedance studies of a green blend polymer electrolyte

based on PVA and Aloe-vera

#### Impedance Studies Of A Green Blend Polymer Electrolyte Based On PVA And Aloe-Vera

S.Selvalakshmi<sup>1</sup>, T.Mathavan<sup>\*1</sup>, N.Vijaya<sup>\*2</sup>, Selvasekarapandian<sup>3,4</sup> M.Premalatha<sup>1,4</sup>,S.Monisha<sup>1,4</sup>

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Abstract. The development of polymer electrolyte materials for energy generating and energy storage devices is a challenge today. A new type of blended green electrolyte based on Poly-vanyl alcohol (PVA) and Aloe-veta has been prepared by solution casting technique. The blending of polymers may lead to the increase in stability due to one polymer portraying itself as a mechanical stiffener and the other as a gelled matrix supported by the other. The prepared blend electrolytes were subjected to Ac impedance studies. It has been found out that the polymer film in which I gm of PVA was dissolved in 40 ml of Aloe-vera extract exhibits highest conductivity and its value is 3.08 × 10<sup>-4</sup> S cm<sup>-1</sup>.

Keywords: PVA, Aloe-vera, Ac impedance

#### INTRODUCTION

lonic conductors are key components for many electrochemical applications, mainly in the field of energy conversion, for example, in photoelectrochemical (PEC) solar cells and fisel cells, in energy storage devices like batteries and in other technological applications like electro-chromic devices, super-capacitors, electrochemical sensors [1]. Several synthetic polymer hosts for solid application have been polymer electrolyte developed in the past, such as poly(ethylene oxide) [2], poly(propylene oxide) [3], poly (acrylonitrile) poly(methyl methacrylate) [5], poly (vinyl chloride) [6], poly(vinylidene fluoride) [7], and poly (vinyl alcohol) [8]. The use of synthetic polymers has drawn much criticism lately due to their complicated synthesis, high costs, and safety issues [9]. To avoid these problems, the interests been shifted over to biodegradable. abundantly available, non-toxic biopolymers or natural polymers. One such natural polymer is aloevera. Pharmaceutical studies like anti-fungal and anti-bacterial studies have been done for aloe-vera. The impedance study of the Aloe-vera leaves has been reported to evaluate the health status of the plant [10]. In the present study, solid blend polymer electrolyte based on PVA and Aloe-vera has been synthesised and imperilled to impedance studies for device application. Blending of polymers is the

most viable approach for the preparation of polymer electrolytes which has been adopted in this research work. Polyvinyl alcohol (PVA) has excellentfilmforming, emulsifying and adhesive pr operties. PVA is non-toxic and has high tensile strength and flexibility. The gel present in the leaf of Aloe-vera has electrical properties and it can generate electricity [11]. PVA having high tensile strength can act as a mechanical suffener and Aloevern extract serves as the gelled matrix in the electrolyte. The prepared green polymer polymer electrolytes were characterized by ac impedance spectroscopy

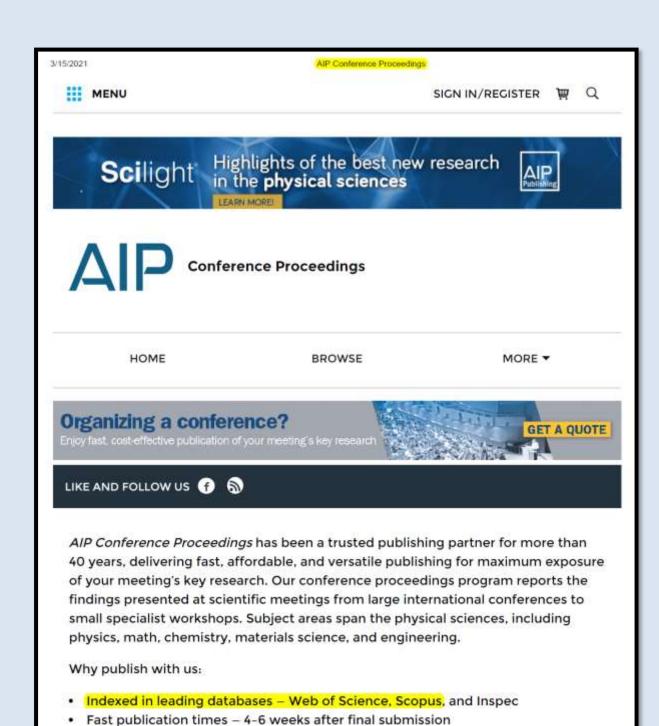
#### Experimental Details:

One gram of Poly vinvl alcohol (PVA) was dissolved in different quantity of Aloe-vera extract to formulate the green blend polymer films. The amount of PVA was kept fixed as 1 gm and the volume of Aloe-vers extract was taken as 10 ml, 20 ml, 30 ml, 40 ml, 50 ml and 60 ml.

The gel present in the leaf of Aloe-vera was detached and was crushed finely in the mixer. The gelatinous solution was double filtered and I gm of PVA was added to the extract. The mixture was magnetically stirred continuously for 36 hours

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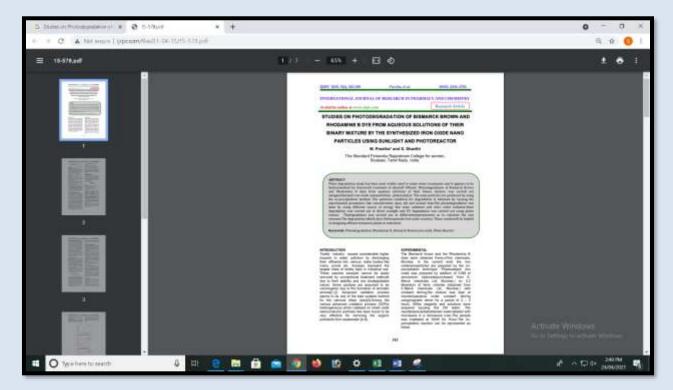
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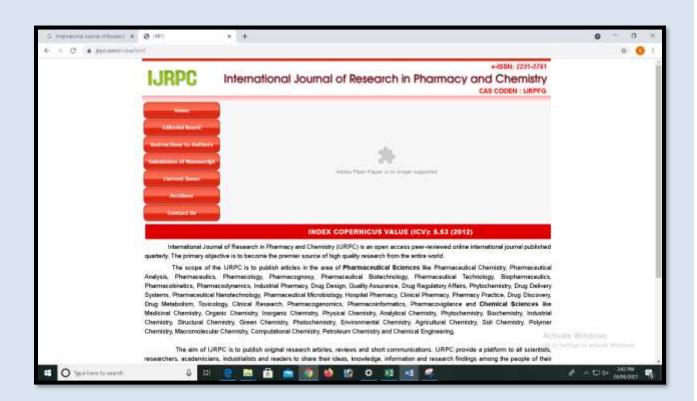
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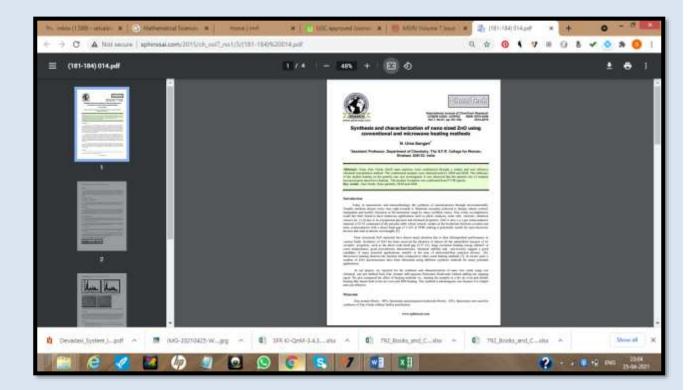


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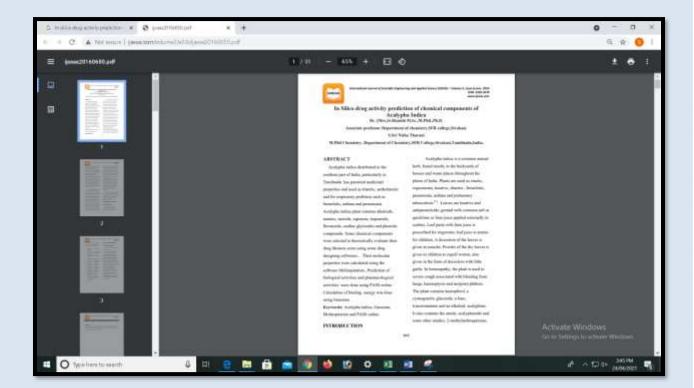


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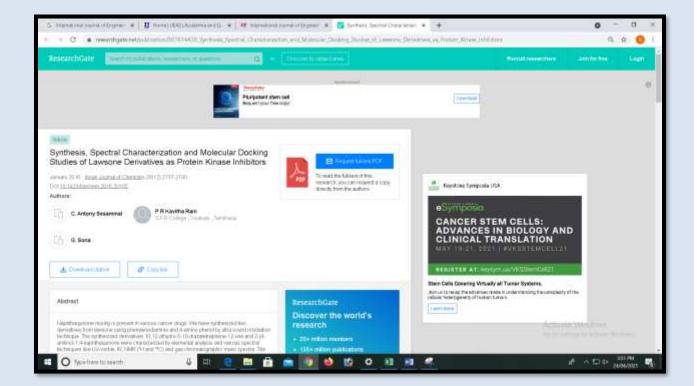


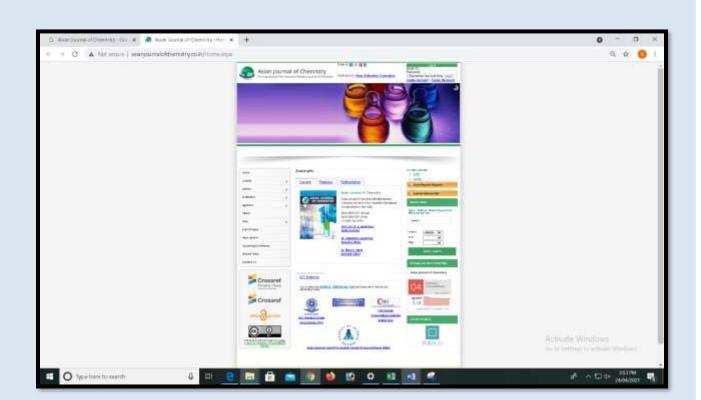
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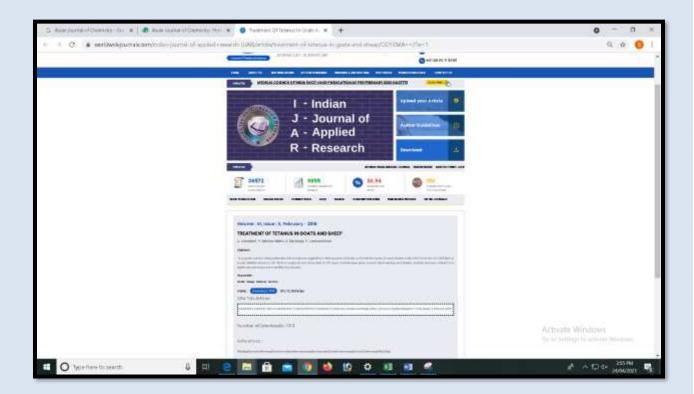




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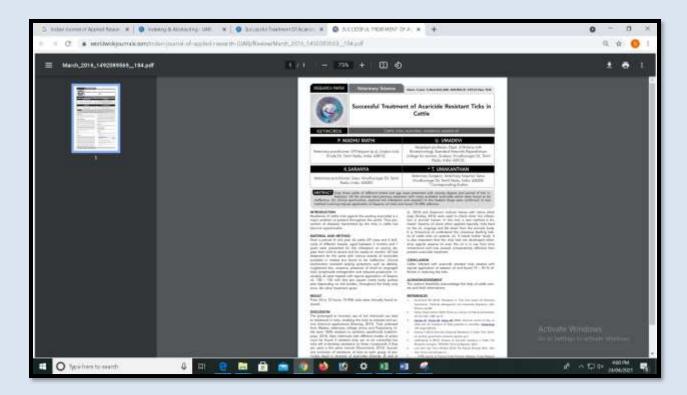
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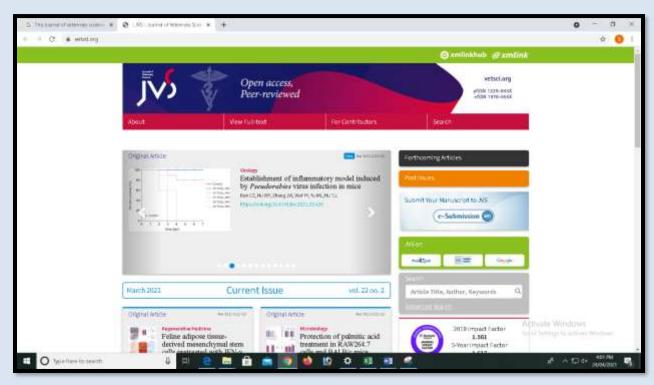
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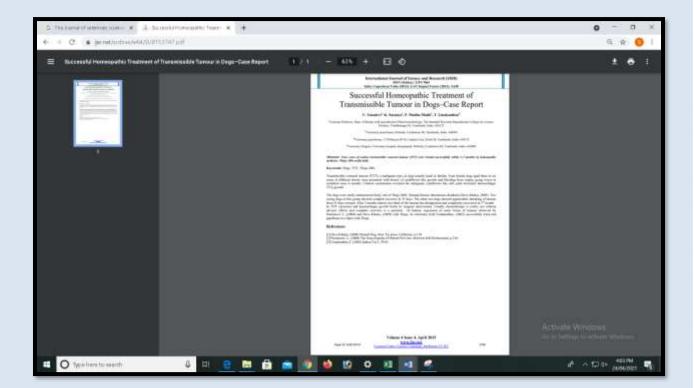


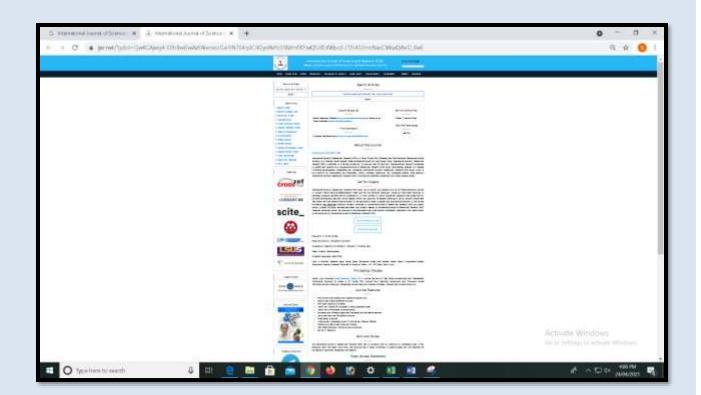
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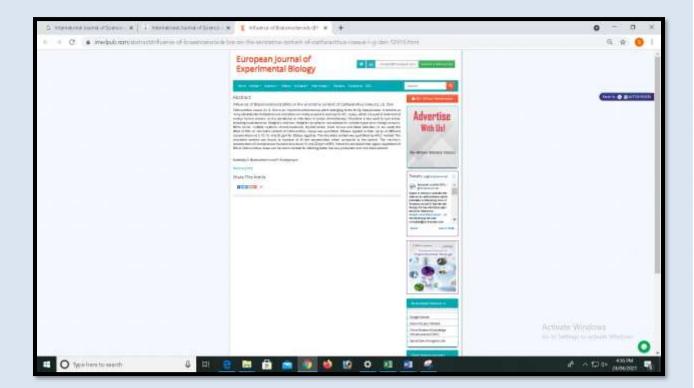


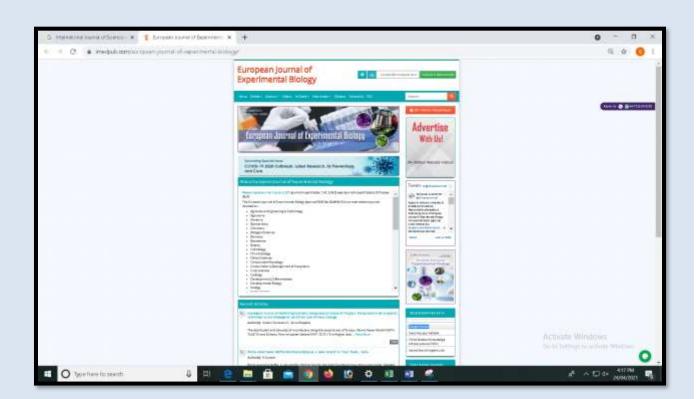
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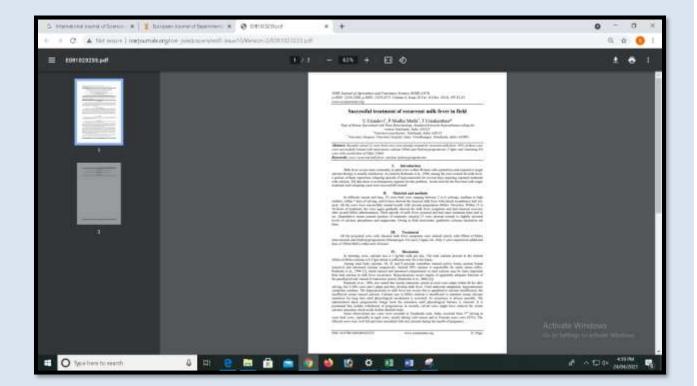


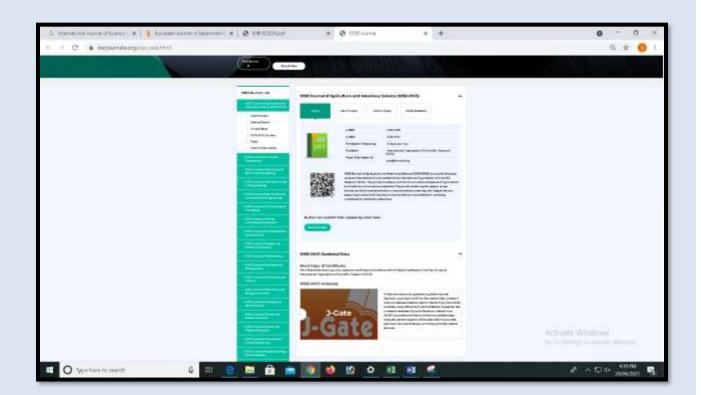


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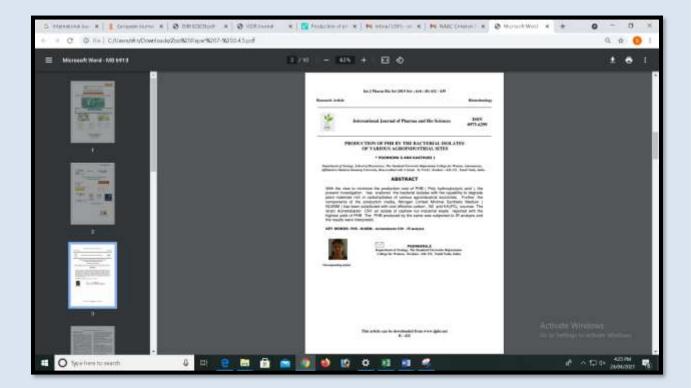


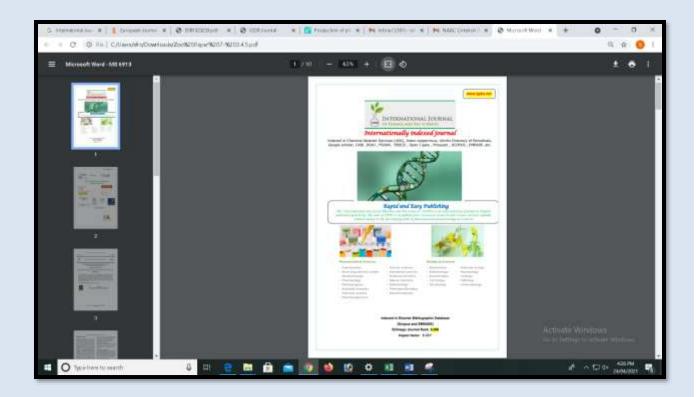
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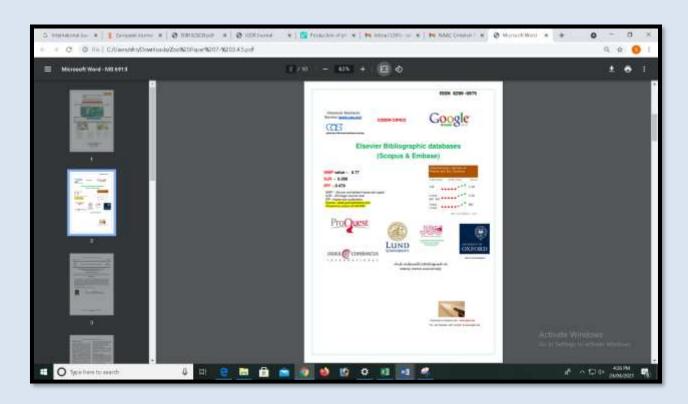
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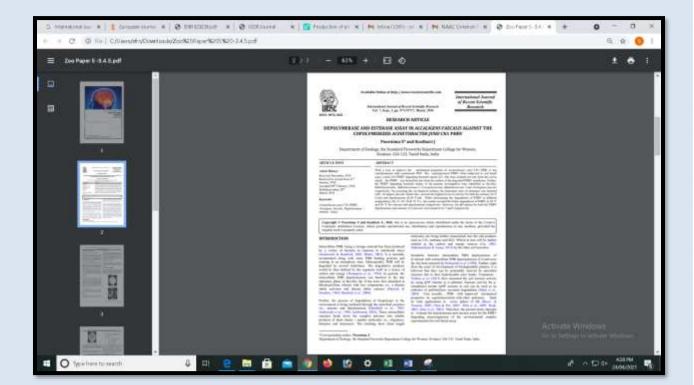


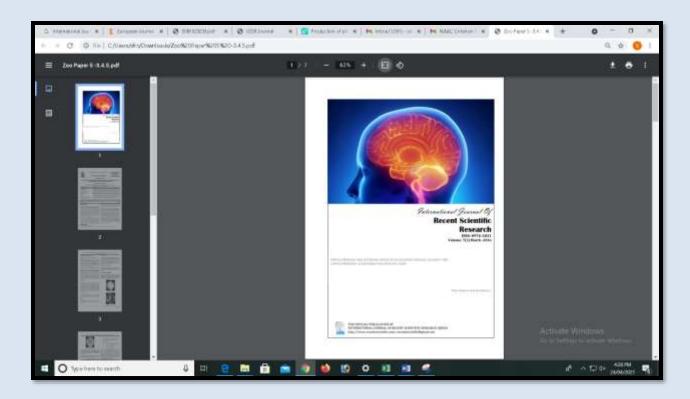
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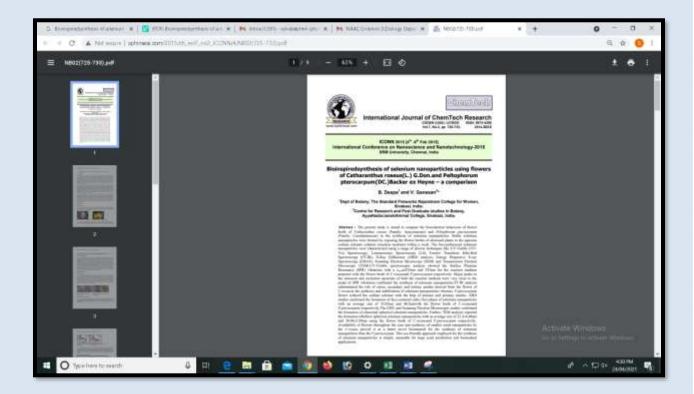
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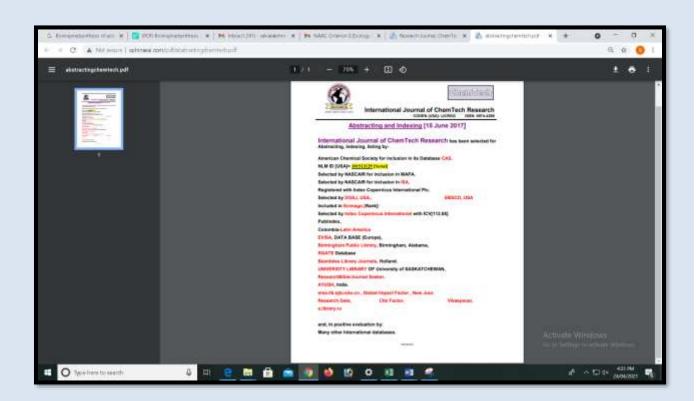
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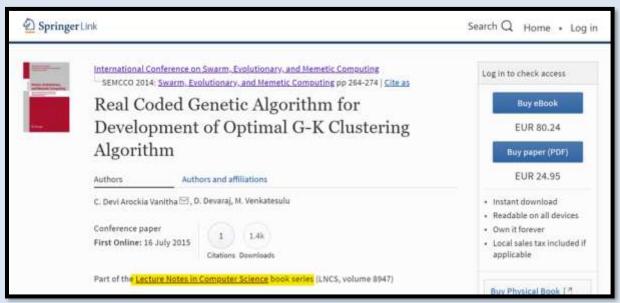


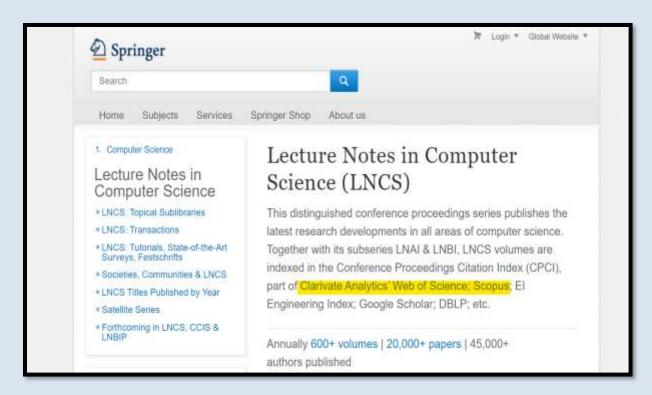
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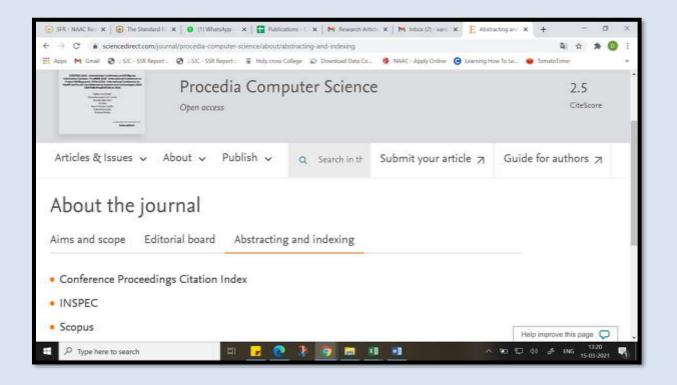
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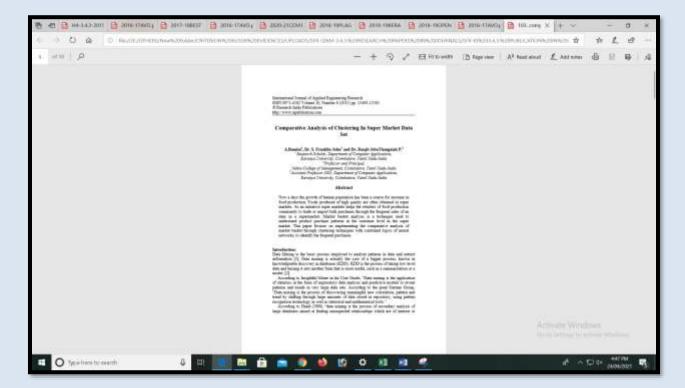


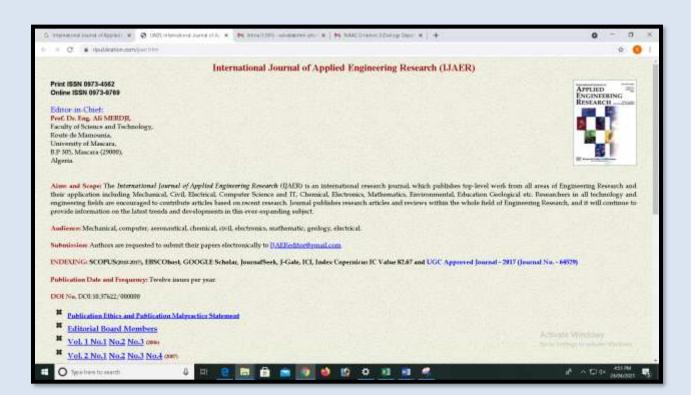
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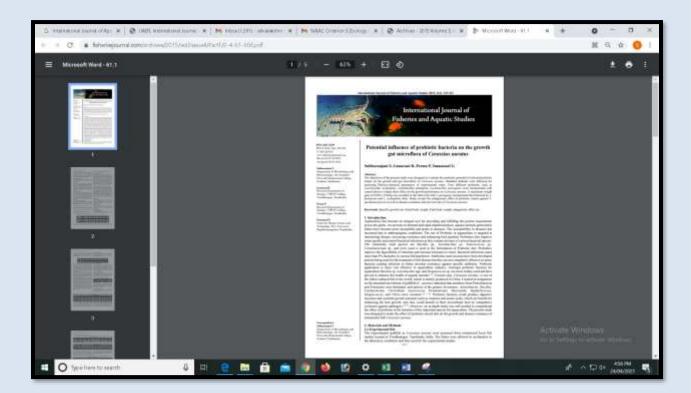


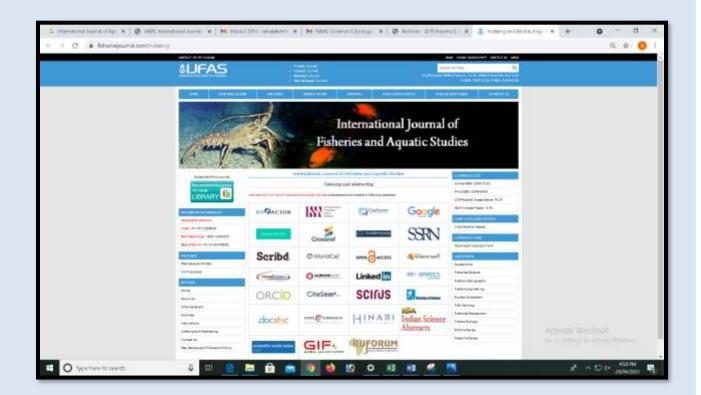
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Name of the Author : Dr.S.Subha Ranjani

Title of the Paper : Potential influence of probiotic bacteria on the growth

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Name of the Author Title of the Paper

: Dr.S.Subha Ranjani

: Supplementation of B. cereus as probiotic in fish feed on

T. trichopterus (Blue Gourami) and calculating its growth

and survival

#### Int.J.Curr.Microbiol.App.Sci (2015) 4(12): 744-751

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#### Original Research Article

Supplementation of B. cereus as Probiotic in Fish Feed of Trichogaster Trichopterus (Blue Gourami) and Calculating its Growth and Survival

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#### ABSTRACT

#### Keywords

Gourami, Bacteriocin, 16s rRNA. Growth rate The study investigated the effect of B. cereus probiotic on the growth performance in T. Trichopterus. B. cereus strain isolated from raw milk was characterized by means of standard biochemical and 16S rRNA gene sequencing studies. Fish feed with and without bacteriocin was prepared and it was fed to the juvenile fishes. After 60 days trial, specific growth rate was increased with the addition of Bacteriocin in fish diets. Likewise the survival rate of juvenile T. trichopterus was 50% for control and 94% the fish diet supplemented with the bacteriocin produced by B. cereus. Thus it was found that in addition of being effective bacteriocin producing B. cereus could also promote the growth of the fish effectively and thus it play an important role in aquaculture nutrition.

#### Introduction

Farming of aquatic organisms like crustaceans, fish, molluses and aquatic plants are termed as Aquaculture and it is identified as aqua farming. Aquaculture is diverged from commercial fishing because the cultivation of fresh water and salt water populations under controlled conditions is former and harvesting of wild fish is later. (ASAP 2009). Ornamental fish are those small sized, live and colourful fish kept in home or public aquaria or in garden pools for recreation. Freshwater ornamental fish contribute 85% of the total global ornamental fish trade (Mohanta2011).

Trichogaster trichopterus (Pallas 1770; Rajesh et al. 2011) commonly called Blue gourami fish, is a common and popular fresh water aquarium fish belonging to the family Belontiidae. An ideal ornamental species should posses attributes like captive survival (including acceptance of artificial diet), attractive colouration pattern, exotically patterned (endemicity), hardiness, peaceful nature, compatible with other species and above all tiny size, so that they can be reared in aquarium throughout their life span. One of the major problems for the growth of ornamental fish farming is the non-





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#### Isolation, Identification and Screening of Cholesterol Degrading Probiotics

K. Vaishnavi, M. Krishma and P. Rajeswari\*

The Standard Fireworks Rajaratnam College for Women, Sivakasi, Tamilnadu \*Corresponding Author E-mail: sholarrpl@gmail.com

#### ABSTRACT

In this study, Cholesterol degrading Lactobacillus was isolated from butter milk sample. The growth performance of Lactobacillus on MRS agar was identified by the colonies were formed due to the utilization of Lactose and by various biochemical tests. Lactobacilli are resistant to various antibiotics and bile salt. Growth performance of selected probiotic in various pH (such as 4, 7 & 9). Temperature (such as 4°C, 37°C, 50°C), and various concentration of Bile salt (such as 0.1g, 0.2g, 0.3g, 0.4g and 0.5g) was detected. In vitro study of cholesterol degradation by probiotic was carried out and it was based on the growth performance and percentage of cholesterol removed from liquid cholesterol. The result of in vitro study reveals Lactobacillus showed the better degradation (21.5%) of cholesterol degraded in liquid cholesterol (at 600 µg/ml).

Key words: Cholesterol, Lactobacillus, Probiotics, Bile salt and MRS medium.

#### INTRODUCTION

Cholesterol is an important basic block for body tissue; elevated blood cholesterol is well-known major risk factor for coronary heart disease<sup>1</sup>. Bile a water-soluble end product of cholesterol in liver is stored and concentrated in the gallbladder and released into the duodenum upon the ingestion of food<sup>2</sup>. Cholesterol being a precursor of bile acid converts its molecule to bile acids replaces those lost<sup>4,5</sup>. Cholesterol is used to synthesis new bile acid in a homeostatic response, resulting in lowering of cholesterol<sup>2</sup>. Many attempts have been made to elucidate the mechanism; one proposed mechanism is the assimilation of the cholesterol by cell wall during growth<sup>3</sup>. Another mechanism is deconjugation of bile salt by bacteria producing bile salt hydrolase.

Lactic acid bacteria capable of lowering blood cholesterol by sticking in the intestinal wall then multiply and producing the enzyme bile salt hydrolase (BSH), which resulted in increased acid conjugated bile that is not easily absorbed from the small intestine compared with bile acid conjugation. Conjugated bile acids by high hence decreasing intestinal absorption of fat and fatty deposits to be reduced.

#### MATERIALS AND METHODS

#### L Sample Collection:

Butter milk was collected from the hostel in the Standard Fireworks Rajaratnam College for Women at Sivakasi to isolate the bacterium Lactobacillus.

Cite this article: Vaishnavi, K., Krishma, M. and Rajeswari, P., Isolation, Identification and Screening of Cholesterol Degrading Probiotics, Int. J. Pure App. Biosci., 3(6): 179-182 (2015). doi: http://dx.doi.org/10.18782/2320-7051.2172

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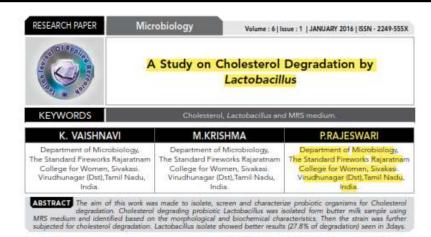
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Name of the Author Title of the Paper

: Mrs.P.Rajeswari

: A study on cholesterol degradation by

Lactobacillus



Cholesterol is a compound belonging to steroid family of molecules (Steinberg, D. 2006). Cholesterol plays major role in human health. Normally, it is need in the body to insulate nerves makes cell membranes and produce some hormones, vitamin D and substance aid for digestion and also need to build healthy cells (HongbaoMa.2004). Excess cholesterol in the blood stream can form plaque in arterial wall (Steinberg, D and Witztum JL. 2002). and plaque build-up causes the arteries to become thicker. and pinque united to be a related to become united, and other vital organs. When too much low-density lipoprotein (LDL) deposits inside the arterial walls where if it is oxidized, it can build-up as hard deposits and cause ath erosclerosis, the disease process that under lies heart attack. Thus the synthesis and utilization of cholesterol must be tightly regulated in order to prevent over accumulation and abnormal deposition with in the body (Fernandez Jes las Heras, L. 2011). Probiotics strains especially lactic acid bacteria have major role to play in the cholesterol level reducing mechanism (Fuller R . 1989).

Lactobacillus bacteria and Bifidobacteria in the host de creased blood cholesterol levels by either decreased total intestinal absorption or removal through solid excretion or by interrupting the entero hepatic cycle of bile acids. Lac-tobacillus bacteria suppressed the re-adsorption of bile acids carrying cholesterol and enhance the removal of m blood through faeces (Hosono, A, 2000). These beneficial organisms are able to deconjugate with These beneficial organisms are able to deconjugate with bile acids such as taturocholic or glycocholic soid. Decon-jugation of bile acid may helps to decrease the serum cho-lesterol in humans, the synthesis of bile acids from choles-terol concentration can reduce the total cholesterol in the body (Buck, M and S.E. Gilliland, 1994).

### Materials and Methods: Sample Collection:

- Butter milk was collected from the hostel in the Standard Fireworks Rajaratnam College for Women, Sivakasi
- to isolate the bacterium Lactobacillus. Serum sample of hypercholesterolemia and hypocho-lesterolemia patients were collected from Clinical Laboratory, Sivakasi

Selective medium was used for the isolation of Lactobacii-lus was MRS medium with the following composition: Pep-tone from casein- 10gm, Yeast extract- 8 gm, D (.) glucose-20gm, Di-potassium hydrogen phosphate- 2 gm, Tween 80- 1 ml, Di-ammonium hydrogen citrate- 2gm, Sodium acetate- 5 gm, Magnesium sulphate- 0.2 gm, Manganese sulphate- 0.04 gm, Distilled water- 1000 ml, pt1- 5.7-2.0.2

#### Identification of Lactobacillus

Identification of Lactobacillus Microscopic observations of the bacterial isolates were studied using Gram Staining and motility tests. [Hanging drop Technique]. Various biochemical tests such as Indole test, Methyl red, Voges Proskauer, Citrate utilization test, Urease test, Oxidase test, Catalase activity and carbohydrate fermentation were carried out for the identification of Lactobacillus

# Cholesterol removal method: (Raghavan et al., 2011) Cholesterol assimilation by using Blood serum choles-

Day - 1
For cholesterol assimilation by Probiotics, 1% of Lactobacillus culture was inoculated into freshly prepared MRS broth, supplemented with bile salt and hyper and hypocholesterplemic patient's serum having cholesterol at various con-centrations such as 180µg/ml, 200µg/ml, 220µg/ml and 240µg/ml respectively. Then the glass vials were inoculated with Lactobacillus culture and anaerobically incubated at 37°C for 24 hours.

Day - 2
The cells were harvested after the incubation period by centrifugation at 10,000 rpm at 4°C for 10 minutes. The cell pellet was washed twice with sterilized distilled water. The cell pellet was suspended in MRS broth containing 0.1 gm of bile salt and patient's serum having cholesterol at various concentrations (180µg/ml, 200µg/ml, 220µg/ml and 240µg/ml) in four vials. This setup was anaerobically incubated at 37°C for 24 hours.

After the incubation period, cholesterol assimilation ability of Lactobacillus to remove the cholesterol: from the media was calculated as percentage from the following equations.

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: Mrs.P.Rajeswari

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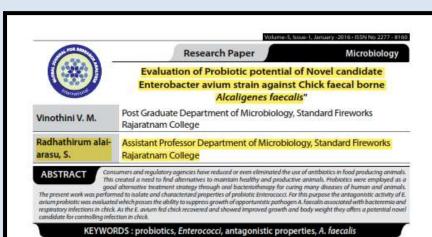


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Name of the Author Title of the Paper

: Dr.S.Radha

: Evaluation of probiotic potential of novel candidate Enternacter avium strain from chick faecalborne Alcaligenes faecalis



#### INTRODUCTION

INTRODUCTION
The Alcodigenes forecalls, an opportunistic pathogen that potentiates viral and other bacterial infection to cause meningitis in new born, bacteremia in cancer patients and associated with purceasit abscess and conneal ulore. Some poults infected with, A. fraccoll's developed mild diamboes, had urate deposits around the closca, were cool to the touch and huddled, and had an odour of the direopings which was characteristic of increased corticosterioid activity. Reports have indicated that Alcolligenes species have also been associated with respitatory infections in the clinick as usually severe and often lethal, and optimal antificiotic therapy is not well established (Omoregie and Osasile, 2012; Moredi et al., 2013). Osagle, 2012; Mordl et al., 2013).

There is a growing demand for problotic functional food and has been used in livestock for decades to decrease the risk of infectious diseases and promote growth performance. Lactic acid bacteria, especially Loctobacillus and Enterococci are the most commonly used microorganisms as problotics have "Generally Recognized As Self" (GRAST status famout et al., 2014; Revocal et al., 2014). Several Safe" (GRAS) status (Ameal et al, 2014; Menconi et al, 2014). Several strategies have been anticipated to identify novel probiotic strains. The properties of probiotic are strain-specific; the selection of strain directly depends on the type of pathogenic infection (Sheela et al, 2010; Gomes et al, 2010). Hence this study intended isolates the strain with high problotic potentiality which may easit in traditional distinct the criteria used for at vitro selection of problotic bacteria, in food preparations, which allow them to be established in the intestinal tract, include ellit tolerance and gastric juice resistance, which enable them to survive and grow to do their impressive action in the gastro-investinal tract (GST). Intestinal tract (GIT).

MATERIALS AND METHODS toolation and identification of Probiotic and Pathogenic strains: The pathogenic strain was isolated from diseased chick fecal waste on MacConRey agar (H media, Mumbal, Indial incubated aerobically at 37°C for 18 to 24 h. The curd samples were collected from college canteen, SFRC, Sinksel was used for isolation of LAB with MRS methum (H media, Mumbal, Indial). Species identification was done by using 105 rDNA sequenting and BLAST analysis. Inhibition assays: For detection of antimicrobial activity, an agar spot test was used (Burthholder and Bhania, 2009). Test cultures for lactabacili was opticed (2-2) µ0 on the surface of MRS agar and incubated for 24 h at 37° C to develop the spots. The agar plates were then overfall with 55 soft agar 0.75% (laW) with pathogens. The plates were incubated at 37° C for 24 h. Zones of inhibition around the central spots were measured.

Bile softs stress resistance: The cultures of appropriate dilution were spread plated on to MRS agar and bile salt agar enriched with 1 % (wt/vol) of bile salt. The plates were incubated 37°C for 24h. After

the colonies grown on MRS agar with that of on MRS agar with bile salt 1 % (wt/vol) and surviving percentage was calculated (Ho et al.

Assessment of Acid tolerance: From the culture of each thain grown overnight cells were centifuged at 5000g for 10 min at 4C. The pellets washed in sterile phosphate buffered salme (PRS) pH 7, and resuspension in the sterile phosphate buffered since (PRS) pH 7, and resuspension it mll containing approximately 10° ciu/ml LAB was transferred into 9 ml phosphate buffered salme (PRS). The pH was adjusted to 1.0, 2.0 and 3.0 using 0.1 N HCl and was incubated at 37°C for 2 and 4h. After different time intervals the acid pH treated samples were inoculated into MHS broth, incubated for 18 h and bacterial growth was determined by measuring the optical density at 600 nm ioD<sub>2</sub>. The values were compared with the control (pH 7) and represented as relative survival percentage.

Auto- and co aggregation assay: The ability of bacteria to auto aggregate and co aggregate was assessed according to the method described by Tareb et al. 2013. The ODO00 of the bacterial suspensteers the state of the state o

Experiment with chick: The chicks (females, 45 days of age) were experiment with orbits: The chicas termaes, 43 casy or age! were used in the present study between February and March, 2013. Two set of domestic chicks were selected for the invivo experiment were obtained from a local hatchery, housed and cared for experiments 1 and.

2. In experiment 1 poults were randomly challenged via oral gavage with approximately 10° chulpoult of pathogenic strain VAF2 and placed in pen. In experiment 2 the poults were given orally only with 10° cfull chick of E. ovuum VEA4 orally. The untreated chick reared for experiment 3 was taken as control. Feeds were given to the poults like rice, grains, etc. In 10 g in each per day. In all experiments chicks were cared and reserd for the period of 40 d. From the days 10 of post infection, the faceal samples were analyzed for LAB and pathogenic stain using MRS and Maconkey selective plating (Karimi Torshizi et al.

Despite of the advances in applied science, poultry diseases con-tinue to cause a considerable economic burden; fortunately, some problems trains have been considered for their positive effects on certain infectious diseases and recently their use to prevent and treat has significantly augmented. Thus there is a clear interest in the identification and characterization of new condidate strains with

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